

This discussion paper is/has been under review for the journal Biogeosciences (BG).  
Please refer to the corresponding final paper in BG if available.

# Chemical composition of modern and fossil Hippopotamid teeth and implications for paleoenvironmental reconstructions and enamel formation: 1. major and minor element variation

G. Brüggmann<sup>1</sup>, J. Krause<sup>2</sup>, T. C. Brachert<sup>3</sup>, O. Kullmer<sup>4</sup>, F. Schrenk<sup>4</sup>,  
I. Ssemmanda<sup>5</sup>, and D. F. Mertz<sup>1</sup>

<sup>1</sup>Institut für Geowissenschaften, Johannes Gutenberg-Universität Mainz, J.J.-Becher-Weg 21,  
55099 Mainz, Germany

<sup>2</sup>Max-Planck-Institut für Chemie, Joh.-Joachim-Becher-Weg 27, 55128 Mainz, Germany

Received: 11 April 2011 – Accepted: 17 May 2011 – Published: 31 May 2011

Correspondence to: G. Brüggmann (bruegmag@uni-mainz.de)

Published by Copernicus Publications on behalf of the European Geosciences Union.

5197

<sup>3</sup>Institut für Geophysik und Geologie, Talstraße 35, Universität Leipzig, 04103 Leipzig, Germany

<sup>4</sup>Forschungsinstitut und Naturmuseum Senckenberg, Senckenberganlage 25,  
60325 Frankfurt, Germany

<sup>5</sup>Geology Department, Makerere University, P.O. Box 7062, Uganda

Received: 11 April 2011 – Accepted: 17 May 2011 – Published: 31 May 2011

Correspondence to: G. Brüggmann (bruegmag@uni-mainz.de)

Published by Copernicus Publications on behalf of the European Geosciences Union.

5198





and Lee-Thorp, 2006; Kohn et al., 1999; Dauphin and Williams, 2004). However, mobilization of an element within the sediment column and eventually its incorporation into the apatite tissue is controlled by environmental and climatic conditions set apart by temperature, redox conditions, pH, and presence of complexing ligands. Therefore, the trace element distribution imposed on apatite tissue during fossil diagenesis can also be used as a proxy for certain environments and climates (Wright et al., 1987; Grandstaff and Terry Jr., 2009; Anderson et al., 2007; Patrick et al., 2004; Trueman et al., 2006).

Most of the studies summarized above used a bulk sample approach in order to determine the chemical and isotopic compositions. Since this technique samples a large volume information is not provided on the spatial-temporal progress of chemical change either during the growth phase of the tooth or during the interaction of the burial environment with the tooth. Such a resolution, however, is essential in order to identify and understand the development of the individual fingerprint of these processes and before detailed paleoecological reconstructions can be made.

This study is part of a comprehensive project which systematically investigates the distribution of major, minor and trace elements as well as stable ( $\delta^{18}\text{O}$  and  $\delta^{13}\text{C}$ ) and radiogenic (Sr, Pb) isotopic compositions in modern and fossil hippopotamid teeth using analytical techniques which provide high spatial resolution, such as the electron microprobe, scanning electron microscope and LA-ICPMS. These techniques can minimize invasion or destruction of fossils and achieve maximum spatial resolution. The present study provides a detailed data base on the composition of tooth apatite in order to identify distinct environments by evaluating the interaction between internal processes during tooth formation, external nutritional controls and diagenetic alteration. We compare tooth material of hippopotamids from different environments in Eastern Africa, representing modern and fossil lakustrine, and modern fluvial environments. We show that the minor element distribution is established during the maturation process of amelogenesis and the patterns of MgO variation to be a tool for distinguishing fluvial, saline and fresh water lake environments. Trace element and radiogenic isotope

5203

variations in the same sample will be presented in a second paper. Stable isotope data have already been published (Brachert et al., 2010) and will be integrated into the results of the major, trace element and radiogenic isotope studies.

## 2 Samples and methods

### 2.1 Samples

Hippopotamids are regarded to be opportunistic grazers living within a small territory due their water-dependency. Samples for chemical analyses of tooth enamel, dentin and cementum were selected from an extensive collection of recent and fossil hippopotamids available to us at the Museum of Natural History (Kampala, Uganda), Institut für Geowissenschaften (Mainz, Germany), Forschungsinstitut Senckenberg (Frankfurt, Germany), and Naturhistorisches Museum Wien (Vienna, Austria). We have decided to study molars of hippopotamids, since these are common fossils of mammals in terrestrial sediments and these robust structures are usually well preserved. In addition, geochemical data in molars are not biased by potential effects of nursing in deciduous teeth. One modern tooth is derived from a zoo collection now housed at the Institut für Geowissenschaften, University of Mainz, Germany (sample S276). Further samples of recent teeth are from animals living in natural environments, including the freshwater lakes of Lake Malawi (Senga Bay, sample Mal-08), and Lake Albert (near the village of Sebugoro in Uganda, sample Alb-10) and the river Nile. The samples from the river Nile represent mini cores of enamel taken at the “Naturhistorisches Museum Wien” in Vienna. Details of the coring procedure are given by Brachert et al. (2010). The animals lived in the Sudanese part of the Upper Nile (samples Nile-1+2), the White Nile (sample Nile-3) and in the Blue Nile (sample Nile-4). The fossil molar teeth studied are from different environments in Eastern Africa, representing fossil lakustrine and fluvial environments, vary in age (recent to 6.5 Ma) and in degree of preservation (Brachert et al., 2010). Sample “Kikorongo” was collected from sediments

5204

of the Kikorongo Crater to the SE of the Rwenzori Mountains. The crater belongs to the Katwe-Kikorongo volcanic field and is believed to be 0.01 Ma old. Today this crater is filled with alkaline, saline water which probably represents a mixture of geothermal and evaporated lake water (Bahati et al., 2005). The remaining 18 fossil specimens were collected in Neogene sediments of the Nkondo-Kaiso and Makondo area at the East shore of Lake Albert (Brachert et al., 2010). Lake Albert is one of a series of large freshwater lakes in the western branch of the East African Rift (Lake Edward, Lake Kivu, Lake Tanganyika, Lake Rukwa, and Lake Malawi) which is a graben system dominantly filled with fluvio-deltaic deposits and lake sediments. The lithostratigraphy as well as the tectonic and paleoecological evolution of Lake Albert is well established (Van Damme and Pickford, 2003; Beuning et al., 1997; Pickford et al., 1993; Senut and Pickford, 1994). Information related to our samples, including the age model, is summarized by Brachert et al. (2010). Taxonomically, all modern teeth studied belong to the species *Hippopotamus ambibius*. However, some of the fossil teeth collected in Uganda are conspicuous for their small size. It has been suggested to classify these using open taxonomy as “Small Hippopotamids” before more detailed systematic analyses have been carried out (Boisserie, 2005).

## 2.2 Microprobe analysis

Carbon-coated polished halves of drill cores and parasagittal or frontal thin sections up to 300  $\mu\text{m}$  thick were used for the microprobe measurements. The samples were analysed for major and minor elements using the Jeol JXA8200 microprobe at the Max-Planck-Institute for Chemistry, Mainz. We used natural minerals and oxides for calibration. Instrument drift during the analysis was corrected, if necessary with repeated measurements of standard apatite (USNM 104021). Table S1 in the Supplement summarizes the measurement conditions, the achieved detection limits and representative standard errors. The spot analyses were done with an acceleration voltage of 15 kV, a probe current of 12 nA and a de-focussed beam with a diameter of 20  $\mu\text{m}$  in order to account for the limited stability of enamel and dentine under the electron beam. Counting

5205

times were 20–60 s on the peak and 10–40 s on the background. Distribution maps for major elements were measured with a dwell time of 110 ms at a beam current of 25 nA at 15 kV with a beam diameter between 1 and 20  $\mu\text{m}$ . Strontium, Cl, F, Ca, and P were detected with WDS spectrometers and Fe, Mg and Si with an EDS spectrometer. The step size varied between 1 and 40  $\mu\text{m}$ .

The major element composition of 20 to 30 spots was systematically determined along 2–3 mm long profiles approximately perpendicular to the enamel-dentin junction (EDJ). In most cases these profiles started at the outside margin of the tooth which is directly exposed to local weathering and leaching, and chemical and/or isotopic interactions with soils, sediments or pore water. The outer part consists either of cement or enamel. The profiles continue crossing the EDJ and ending with a few spot analyses in dentin (Fig. 1). If possible, two profiles were sampled near the apex and cervix, respectively, to cover the full spectrum of compositional variability. In a few specimen profiles were measured from the cervix to the apex, close to and parallel to the EDJ (S276-L, Mal-08-L, 5306-L).

## 3 Results: major and minor element composition of hippopotamid molar teeth

Tables 1, 2, and 3 summarize the composition of enamel, dentin and cement from the modern specimens and of fossil teeth from Lake Kikorongo and Lake Albert. The complete set of 1880 major element analyses on 26 teeth specimen is presented in the electronic attachment (Tables S2–S4). One concern when analyzing different tooth materials with relative large spot sizes is their variable porosity which pretends different apatite compositions. The total oxide sum in the various segments of the teeth is always considerably less than 100 wt. % ranging from about 93 wt. % for enamel to 70 wt. % for dentin and cement (Total in Tables 1, 2, 3). The deviation from 100 % is mainly due to the presence of carbonate and residual organic substances – which have not been determined – and of empty pore spaces within the measured spot area. This also becomes obvious in element maps like those of Ca shown in Fig. 2. It suggests a large concentration difference between dentin, cement and enamel although effectively

5206

the difference is just about 3 wt. % CaO. The Ca distribution also indicates an intensity decrease from the apex towards the cervix (Fig. 2). These variations do not only indicate systematic differences in the Ca content of bioapatite, but also reflect differences in mineral density between enamel and dentin (~95 wt. % versus 70 wt. %, respectively; Pasteris et al. (2008) and within enamel, which is characteristic of mammal enamel in general (Robinson et al., 1995; Weatherell et al., 1974). In order to facilitate compositional comparison between different tooth materials the data shown in Tables 1 to 3 and in the Figs. 3, 4, and 8 are normalized to 100 %. The original analytical results are given in the electronic attachment. In our sample set the discussion of concentrations or element ratios using CaO or P<sub>2</sub>O<sub>5</sub> as the denominator are interchangeable and provide equivalent interpretations and conclusions regarding chemical variations induced by metabolic/nutritional and diagenetic processes.

The concentrations of the main oxides CaO and P<sub>2</sub>O<sub>5</sub> in modern and fossil enamel vary by about 1 % (1 standard deviation; Table 1), although dentin and cement show larger concentration variations of up to 7.6 % (Tables 2, 3). Typically, fossil dentin and cement are more heterogeneous and have higher CaO and lower P<sub>2</sub>O<sub>5</sub> contents than their modern counterparts (Fig. 3). In fossil samples dentin and cement have lower CaO and P<sub>2</sub>O<sub>5</sub> contents than enamel (Fig. 3). Modern and fossil enamel have on average similar CaO and P<sub>2</sub>O<sub>5</sub> contents. This is also reflected in the similar CaO/P<sub>2</sub>O<sub>5</sub> ratio of modern and fossil enamel ( $1.28 \pm 0.3$  versus  $1.31 \pm 0.2$ , respectively; Table 1). Modern dentin and cement have lower and higher whereas fossil dentin and cement have always higher CaO/P<sub>2</sub>O<sub>5</sub> ratios than enamel (Tables 2, 3). Figure 4 summarizes typical distribution patterns along profiles from the outside rim, across the EDJ into the dentin. Enamel thickness has been normalized to 1 in order to minimize the influence of variable enamel thickness in different specimen. The concentration patterns of CaO and P<sub>2</sub>O<sub>5</sub> across the enamel are flat and similar for all specimens, reflecting the small standard deviation of the average concentrations. The P<sub>2</sub>O<sub>5</sub> contents in dentin and cement are systematically lower, but the CaO can be enriched or depleted relatively to the enamel (Fig. 4).

5207

The third most abundant component in modern teeth is Na<sub>2</sub>O (~1 wt. %). Modern tooth materials and fossil enamel have similar concentrations, whereas fossil dentin and cement are on average lower in Na<sub>2</sub>O (Tables 1–3, Fig. 3). In element maps, the contacts of cement/enamel and enamel/dentin are clearly delineated by sharp concentrations changes (Figs. 2, 4). Even within the enamel Na<sub>2</sub>O contents increase by about 40 % from the outside towards the EDJ (Fig. 2). This is observed in all profiles perpendicular to the EDJ throughout the tooth crown, regardless of provenance and geological age of the specimen and is not due to changes in mineral density but reflects variable apatite composition. Similar Na<sub>2</sub>O distributions in enamel have been observed in other mammal teeth, for example from rats, porcupine and humans (Norén et al., 1983; Driessens et al., 1990).

The distribution pattern of MgO in modern and fossil Hippopotamid enamel is similar to that of Na<sub>2</sub>O, as MgO concentrations increase from the enamel rim towards the EDJ, although the enrichment is slightly less (~35 %; Figs. 2, 4). However, it differs in that modern enamel has 4 to 8 times lower concentration than cement and dentin (Tables 1, 2, 3, Fig. 3). This pattern of MgO distribution between different materials is also typical for many other mammal teeth (Driessens et al., 1990; Steinfert et al., 1991). Fossil enamel from Lake Albert has MgO contents of about 0.22 which is within the range of the MgO contents in modern enamel (Fig. 3). There are two obvious exceptions; enamel of the hippopotamus tooth from Lake Kikorongo and enamel from the White Nile River have distinctively higher MgO contents (~0.4 wt. %), and higher MgO/P<sub>2</sub>O<sub>5</sub> (<0.07) and MgO/Na<sub>2</sub>O ratios (>0.28) than all the remaining enamel samples (Table 1).

Chlorine content in modern and fossil enamel is overlapping (~0.4 wt. % on average) and it is always higher than that in dentin and cement (Tables 1–3, Figs. 2, 3). However, the average Cl content in enamel varies by about 40 %. This rather large variation does not reflect differences among different teeth but is mainly due to intra enamel variation where concentrations may increase from 0.1 to 0.7 wt. % Cl as one moves from the EDJ towards the outer rim (Fig. 4). This distribution can be observed all along the

5208

tooth crown (Fig. 2). The study of Kohn et al. (1999) describes a similar concentration range in different regions of enamel from fossil hippopotamids and studies on other mammal teeth corroborate these observations (Norén et al., 1983; Driessens et al., 1990).

5 Modern tooth material has low F contents (0.1–0.7 wt. %; Tables 1, 2, 3; Fig. 3) and it represents with regard to its halogen abundances (Cl, F, OH) the composition of hydroxyapatite. Fossil material has systematically higher F contents than the modern examples. This is particularly true for fossil cement and dentin which are enriched by containing on average about 2 wt. % of F (Fig. 3). This concentration contrast between enamel and cement and dentin is clearly seen along profiles (Fig. 4). Within 10 the enamel, the distribution of F differs among modern and fossil material. In modern enamel, F contents stay rather constant or decrease from the EDJ towards the outside. Previous studies on human, monkey, rat or porcupine enamel, however, observed the opposite trend (Norén et al., 1983; Driessens et al., 1990). Such a distribution pattern 15 is displayed in most of our fossil enamel material (Fig. 4); however, we also find an U-shape distribution where the central enamel region has the lowest F contents (Fig. 4). Even in these cases the outer rim has always the highest F contents.

In modern specimens concentration of the remaining analyzed components (SrO, FeO, MnO, K<sub>2</sub>O, SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, and SO<sub>3</sub>) are consistently very low, often close to or 20 below the detection limits, so that the relationships between the different tooth regions cannot be reasonably well resolved (Tables 1, 2, 3). Whereas modern tooth parts and fossil enamel have on average similar SrO contents, the concentrations in fossil dentin and cement are higher and more variable (Fig. 3). One specimen stands out (Lake Kikorongo) because it has consistently higher SrO contents, which are 3 times 25 higher in enamel compared to all other samples (Table 1; Fig. 4). Fossil materials, in particular dentin and cement, are highly enriched in FeO and MnO and show large variations of concentrations (>50 %; Tables 1, 2, 3; Fig. 3). Variable FeO contents are easily recognized macroscopically by the different coloring of the teeth samples (Fig. 1). Modern tooth material is entirely white and concentrations of FeO and MnO are below

5209

0.1 wt. % (Tables 1, 2, 3). Fossil dentin and cement and to a lesser degree enamel are characterized by light to dark brown staining and may contain up to 24 wt. % FeO and 1.4 wt. % MnO (sample 053-06; Table 2 and Fig. 4). The FeO distribution in fossil enamel is systematic in that concentration can increase from detection limit at the EDJ 5 towards wt. % levels at the enamel rim (Fig. 4).

The very low concentrations of K<sub>2</sub>O, SiO<sub>2</sub>, and Al<sub>2</sub>O<sub>3</sub> in modern and fossil materials indicate that crystallization or mechanical introduction of silicate phases into the fossil during diagenesis is not of importance. These components will not be discussed further on.

#### 10 4 Discussion

Teeth are composite materials consisting of organic and inorganic components which undergo profound changes in chemistry and structure during growth and eventually during fossil diagenesis (Boskey, 2007; Kohn et al., 1999; Smith, 1998; Robinson et al., 1995). In the following, we decipher the chemical signatures preset during the 15 growth of the tooth and later on superimposed during fossilization. This is done by comparing the chemical composition of modern and fossil material on the basis of the bulk composition as well as on the element distribution observed across enamel and dentin growth structures. These comparisons are supported by statistical methods such as linear regression analysis and single factor analysis of variance (ANOVA). 20 However, the compositions of dentin and cement are highly variable. ANOVA does not reveal significant differences among tooth populations and no systematic concentration relationships between these materials can be observed.

In order to generalize the observation made on the distribution of the major elements and to compare the compositions of enamel of different geological age and habitat, the relationship between concentration and enamel thickness ( $x$ ) are quantified by linear 25 regression analysis for each EDJ-rim profile in each specimen (number of profiles: 18 modern, 42 fossil). In order to account for variable thickness of enamel in large and





along many fossil enamel profiles shows a U-shape pattern. Although this pattern may contain a primary fingerprint, it appears to be dominated by secondary F addition through diffusive migration of the pore fluids into the tooth. Such processes have been discussed by Millard and Hedges (1996) and Kohn (2008) in order to explain similar distribution patterns in fossil bones and teeth. Analogous to the Fe trend F enrichment occurs without changing the P/Ca ratio (Fig. 5c, d). This implies that no other Ca or P-bearing phase, such as fluorite ( $\text{CaF}_2$ ) or vivianite ( $\text{Fe}_3^{2+}(\text{PO}_4)_2 \cdot 8(\text{H}_2\text{O})$ ), has been introduced or crystallized during fossilization and diagenesis. Thus, low  $\text{P}_2\text{O}_5$  and CaO contents in fossil enamel are solely due to their "dilution" by other components such as F and FeO.

The  $\text{SO}_3$  concentration in teeth material is highly variable (Fig. 3) and ANOVA cannot distinguish teeth populations (Table 5). However,  $\text{SO}_3$  concentrations in fossil enamel tend to be higher than in modern enamel (Fig. 3; Table 4) and there is a significant correlation between  $\text{SO}_3$  and F contents in fossil enamel ( $R^2 = 0.43$ ,  $p < 0.0001$ ), but not in modern material. The distribution pattern of F and  $\text{SO}_3$  along profiles of fossil specimens is also similar (Fig. 4), although no overall trend can be recognized (Table 4). This suggests that the  $\text{SO}_3$  content and distribution in the teeth has also been significantly modified during diagenesis.

The FeO, F and  $\text{SO}_3$  contents do not systematically change with the geological age of the hippopotamid teeth. This implies that local sedimentary factors influencing the mobilization of these components, such as rock composition, pH value and redox potential, control the introduction of these elements into the tooth materials during fossilization, rather than chemical exchange on large time scales. With regard to the FeO content of fossil samples, those from Lake Kikorongo appear to be distinct because they have much lower contents of FeO, F and  $\text{SO}_3$  than fossil teeth from Lake Albert (Table 4). This difference probably reflects the depositional environment in which the teeth have been fossilized. Throughout its history Lake Albert has been a freshwater lake (Pickford et al., 1993). In contrast, the water of Lake Kikorongo is saline and alkaline representing an environment where FeO mobilization is limited. Hence,

5213

although these elements bear little information on the diet of primeval hippopotamids, post-mortem concentration changes in fossil teeth may provide information on past ecological changes, because the mobility of the element in the local pore water is affected by climate parameters such as temperature or precipitation.

#### 5.2 Chemical characteristics of dentin and cement from Hippopotamid teeth

Dentin and cement exhibit a rather "spongy" texture and for this reason these materials become severely altered during diagenesis. This has been clearly documented for elements like F and Fe (see discussion in previous chapter). However, the influence of diagenesis cannot be unambiguously evaluated for major and minor components ( $\text{CaO}$ ,  $\text{Na}_2\text{O}$ ,  $\text{MgO}$ ,  $\text{Cl}$ ,  $\text{P}_2\text{O}_5$ ) because we have not systematically investigated the composition of dentin and cement along specific growth structures. The composition of dentin and cement differ from enamel even among modern teeth which becomes obvious along profiles from the outer margin of the tooth crown towards and into the dentin (Fig. 4). The interface between enamel and dentin or enamel and cement is narrow ( $<100\ \mu\text{m}$ ) and well defined for most elements. The pattern of concentration change across the dentin-enamel and cement-enamel has the same trend for individual elements, although with different magnitudes. Dentin and cement are depleted in CaO and Cl, but are enriched in MgO and  $\text{P}_2\text{O}_5$  relative to enamel (Fig. 4). One exception appears to be  $\text{Na}_2\text{O}$ , which is depleted in dentin but enriched in cement relative to enamel. However, there are no systematic variations along the profiles within dentin and cement which appear to be inhomogeneous on the scale of ten's of micrometers. The low total sum of the chemical analysis in dentin and cement compared to enamel (Fig. 4) reflects their lower mineral proportion relative to the organic matrices. This, however, does not explain the variable  $\text{CaO}/\text{P}_2\text{O}_5$  ratios as indicated by the atomic proportions of the elements (Fig. 6) and which differ from that of enamel.

Although there are large concentration variations in and between modern dentin and cement, the relative atomic proportions of the main components are very similar and in these terms fundamental chemical/mineralogical differences between these

5214

two materials cannot be deciphered. The Na variation is relatively small (Fig. 6) and comparable to that in enamel which indicates that this element is very well controlled by metabolic processes. However, Cl and Mg in dentin and cement vary substantially and appear to delineate different habitats. For example, cement and dentin from Lake Malawi Hippopotamid are enriched in Mg, whereas the material from the zoo animal is enriched in Cl relative to the other samples (Fig. 6). The Na/Ca ratio and the variation of Mg/Na ratio in modern cement and dentin is similar to that of enamel and distinguishes different environments (Fig. 6). Nonetheless, the pattern displayed by this material is different from that of enamel. Whereas enamel from the Nile River Hippopotamid has a distinctively high Na/Mg and the zoo enamel has the lowest Na/Mg ratio, dentin and cement from the zoo animal has the highest Na/Mg ratio (compare Figs. 6d and 7d). These differences are difficult to reconcile. One plausible explanation is that the timing of the mineralization of dentin and cement differs from that of enamel which may result in different chemical signatures, perhaps further accentuated by changing environmental and nutritional factors.

The range of minor element concentrations in fossil dentin and cement overlaps with and in cases extends that of modern specimen (Fig. 6). The proportions of Na, Mg and Cl in the specimen of Lake Kikorongo are higher than those in fossils from Lake Albert. However, the contents of these elements in material from Lake Kikorongo overlap with those in the modern specimen, which suggests that they do not necessarily reflect element enrichments indicative for a saline ecosystem. The concentration in the fossil material from Lake Albert, on the other side, has lower concentration than the modern one. In particular, the low Na proportion in fossils from Lake Albert is reflected in the high Ca/Na ratio shown in Fig. 6h, whereas materials from Lake Kikorongo plot along the line defined by enamel and modern cement and dentin. Thus, a reasonable explanation is that the fossil material lost Mg, Na and Cl during fossilization.

5215

### 4.3 Chemical characteristics of Hippopotamid tooth enamel: recognizing the in-vivo control on element distribution

The distribution of the principal components ( $P_2O_5$ , CaO, MgO,  $Na_2O$ , Cl) is similar in fossil and modern enamel and is little disturbed by secondary processes. For example, the slopes calculated with the regression analyses are similar for individual elements regardless of the origin and the geological age of the tooth (Fig. 4; Table 4). In some cases minor enrichment or depletion of these elements exists near the outer enamel rim (Fig. 4) indicating minor mobilization during diagenesis. In contrast, near the EDJ concentration variations are small (Table 4). The EDJ is well defined showing no gradual concentration change into the dentin (Fig. 4) which implies very limited chemical exchange or migration by diffusion between enamel and dentin. Thus, the most preferred sampling site for chemical analysis, if focusing on the unaltered, primary fingerprint, is enamel close to the EDJ (i.e.  $x = 1$  in Table 4).

In particular the relative atomic proportions of the main components are indistinguishable in fossil and modern enamel (Fig. 7). The molar P/Ca ratio in modern enamel is overall well constrained and varies within tight limits ( $0.618 \pm 0.018$ ; Fig. 7) showing no clear dependency with tooth growth. Fossil enamel displays the same P/Ca variation ( $0.604 \pm 0.012$ ) and ANOVA could not identify significant differences ( $p < 0.05$ ) among enamel populations (Table 5). Similarly, the ranges of the atomic proportions of Mg, Na and Cl in enamel from fossil samples overlaps with that of modern enamel (Fig. 7). The overall limited variation in major and minor elements is also confirmed by ANOVA, which cannot distinguish different enamel populations by means of the Total of all measured components (Table 5). This Total decreases from the outer margin towards the EDJ in fossil and modern enamel to similar degrees ( $\sim 2.5$  wt. %; total in Fig. 4 and Tables 1 to 5). One important chemical component not included in our measurement is the carbonate content of the enamel. Considering data from other mammals, carbonate shows the opposite behavior, decreasing from the EDJ towards the outside (Aoba and Moreno, 1990; Verbeeck et al., 1985; Zazzo et al., 2005) which would compensate

5216

the variation of the Total we measured. Therefore, diagenesis or post-mortem recrystallization has not significantly altered the major components of fossil enamel, which might also rationalize why oxygen or carbon isotopic studies on enamel are most successful tools in revealing paleoclimate and -environmental changes.

5 The variation of the main components along the profiles follows tree patterns. CaO and P<sub>2</sub>O<sub>5</sub> show flat trends along the profiles from all specimens and therefore the concentrations at the inner and outer enamel rims overlap, within analytical error (Table 4). Na<sub>2</sub>O and MgO display always positive slopes from the enamel rim towards the EDJ, whereas Cl always follows a negative trend. These three components have the highest  $R^2$  factor among all studied elements (Table 4), which implies that during enamel growth their distribution is well controlled and follows the same laws in fossil as well as in modern animals. Ratios such as Na/P, Na/Ca, Mg/P, or Mg/Ca are not constant but systematically increase along constant Ca/P ratios from the outer rim towards the EDJ (Fig. 7). This means that in order to draw meaningful conclusions about changing nutritional intake, equivalent sample locations within the enamel body have to be compared.

10 In summary, the compositional variations of P<sub>2</sub>O<sub>5</sub> and CaO are decoupled from the variation of Na<sub>2</sub>O, MgO or Cl which implies different control mechanism during the process of enamel formation. In addition, there is the remarkable feature that the general concentration trends of Na<sub>2</sub>O, MgO or Cl across different enamel profiles are similar, but the concentrations of the three elements follow different slopes and the variations from the enamel rim to the EDJ are large ranging from about 40 % for MgO and Na<sub>2</sub>O to more than 200 % for Cl (Table 4). Hence, two principle questions arise: i. what controls the different slopes of the element distribution along the enamel profile, and ii. what determines the degree of enrichment of MgO and Na<sub>2</sub>O or depletion of Cl during the formation of enamel?

5217

### 4.3.1 Recognizing discrete sedimentary environments

Abrupt chemical changes along the enamel profiles have never been observed in our specimen. This implies that major changes of diet have not occurred and that individual animals did not moved to substantially different habitats during the time of enamel formation. This had been anticipated considering the autochthonous behavior of hippopotamuses. The small variation of average concentration of CaO, P<sub>2</sub>O<sub>5</sub> and the tightly defined distribution pattern ( $R^2 > 0.8$ ) with similar Cl and Na<sub>2</sub>O variations across all profiles (Figs. 4, 6) imply a tight physiological control of the concentration of these elements. Overall, the Na/Ca ratio appears to be the least variable, because fossil and modern enamel follow equal trends of constant Na/Ca (Fig. 7 d, h). Biological extracellular fluids, such as blood plasma and enamel fluid, are in essence NaCl solutions (Aoba and Moreno, 1987; Simmer and Fincham, 1995; Larsson et al., 1988; Fraústo da Silva and Williams, 2001) and may render compositional changes induced by food intake hard to exist. However, ANOVA and the Scheffé tests suggest the Na<sub>2</sub>O variation in the population of fossil enamel from Lake Albert to be distinct from those of Lake Kikorongo and modern environments (Table 5). This distinction reflects the low Na<sub>2</sub>O content in fossil enamel from Lake Albert (Table 4). The Scheffé tests also indicate a significant difference between Cl contents in modern and fossil enamel (Table 5) and this is likely due to the low Cl contents in fossil enamel relative to modern enamel (Table 4). It is not apparent whether these population differences in Na<sub>2</sub>O and Cl represent a primary feature induced by a diverse diet or by vital effects. However, it should not be related to loss of Na<sub>2</sub>O and Cl during alteration. The distributions of these two elements are highly regular in all teeth; this would not be expected, considering the different geological ages and environmental settings, if these elements became mobilized during alteration processes.

25 The larger variation of MgO ( $R^2 = 0.58$ ) along the regression lines, and the variable MgO/Na<sub>2</sub>O ratios in enamel from different teeth suggest less physiologically control than in the Na-Cl system (Fig. 7; Table 4). Indeed, enamel from single locations

5218

has rather well defined MgO/Na<sub>2</sub>O ratios and there are differences which demarcate sample origins. ANOVA corroborates and the Scheffé test confirms the distinct MgO distribution among the different enamel populations (Table 5). The fossil specimens are from two different environments: the saline Lake Kikorongo which is of volcanic origin (Bahati et al., 2005) and the Neogene freshwater equivalent to Lake Albert (Lake Obweruka) (Brachert et al., 2010; Van Damme and Pickford, 2003). Compared to the fresh water lakes saline waters are characterized by their high concentrations of Na and Cl but also of Sr and MgO (Hammer, 1986). Enamel from the tooth of the Kikorongo specimen has distinctively higher SrO and MgO concentration than the Lake Albert samples (Figs. 4, 7; Tables 1, 5). In particular the relative proportion of Na-Mg-Ca or the MgO/Na<sub>2</sub>O ratio can readily distinguish these two environments (Fig. 7h). Thus, the concentrations of these elements in enamel have the potential to be used as proxies distinguishing saline lake from fresh water lake environments. It is interesting to note that the MgO/CaO ratio increases in fresh water lakes if prolonged evaporation causes the precipitation of calcite (Crausbay et al., 2006; Yu et al., 2002). Similarly, pedogenic carbonate (calcite) forms in dry climates possibly increasing the MgO/CaO ratio in fluids available to plant roots. There is no significant variation with age in the MgO/Na<sub>2</sub>O ratio in fossil enamel from Lake Albert suggesting that the lake water did not experience major compositional changes between 7 and 1.5 Ma. This is supported by the sediment record which is dominated by clastic, arenitic to pelitic sediments without any significant amounts of carbonate and only minor limestone beds (Pickford et al., 1993). However, oxygen stable isotope data indicate a trend of continual evaporation of the lake without precipitating carbonate from about 7 to 2 Ma and carbon isotope indicate a vegetation change from C3 to C4 plant somewhere 4 and 2 Ma (Brachert et al., 2010). Obviously, the stable isotope systems are more sensitive to evaporation than MgO/CaO ratios.

Modern enamel from hippopotamids living in the Upper and Blue Nile Rivers have systematically lower Mg/Na ratios than the remaining samples from the White Nile River, Lake Albert, Lake Malawi, or from the zoo (Figs. 7d, 8). Even enamel from

5219

hippopotamids living in different parts of the river Nile system can be distinguished because the MgO/Na<sub>2</sub>O and the MgO/CaO ratios in White Nile enamel is higher than that from the Blue Nile and Upper Nile (Table 1; Figs. 7, 8). It is well established that the chemical and isotope composition of the water and the sediment load in the Blue Nile and White Nile Rivers differ significantly reflecting the geologically different catchment areas in volcanic rocks dominating the Ethiopian Highland and the Precambrian basement in Uganda, respectively (Kempe, 1993; Revel et al., 2010; Scrivner et al., 2004; Talbot et al., 2000; Palmer and Edmond, 1989). In addition, the White Nile River suffers from strong evaporation and passes several swampy and saline areas (Sudd, Nuba and Gezira zone) before joining the Blue Nile River (Salama and Selley, 1997). This could explain the higher MgO contents and MgO/Na<sub>2</sub>O ratios in enamel from the White Nile (Fig. 7). Enamel from Blue Nile and Upper Nile show similar MgO variations (Fig. 7). This probably reflects the dominant contribution of the Blue Nile to the main Nile, with regard to the amount of water flow and sediment load (Foucault and Stanley, 1989; Williams, 2009). The compositions of enamel from animals which lived in the modern, large freshwater lakes (Lake Albert, Lake Malawi) are overlapping (Fig. 8). In fact, no significant difference should be present in enamel from hippopotamids from these two lakes, when their similar lake character, the similar climate and their similar country rock assemblage is taken into account. Therefore, the two branches of the Nile river system and the shallow and deep lakes, respectively, define distinct settings which exert a strong imprint on chemical signatures of hippopotamid enamel via nutritional uptake (food and drinking water).

#### 4.3.2 Implications for the formation of enamel apatite

Previous studies mainly performed on human, rat or porcupine teeth have established that enamel formation is the result of coordinated extracellular processes involving organic matrix molecules and inorganic ion fluxes that regulate the nucleation and growth of developing apatite crystals (Simmer et al., 2010). During the initial secretory stage of enamel formation, ameloblasts form a protein matrix which modulates

5220

the crystallization of long thin ribbons of enamel crystals until the full enamel size is reached (Smith, 1998; Margolis et al., 2006; Simmer et al., 2010). Subsequently, during the maturation stage, these crystals grow in thickness and width maintaining the structural organization of the enamel established during the secretory stage (Robinson et al., 1995). This phase is accompanied by the almost complete removal of proteins and pore fluid, becoming replaced by bioapatite. Thus amelogenesis occurs in a confined space separated from the interstitial body fluid by an epithelium of ameloblasts, which control the in- and outflow of organic and inorganic molecules and ions at each step of enamel formation (Hubbard, 2000; Smith, 1998). Even the regressing epithelium layer during maturation keeps the maturing enamel isolated from the surrounding connective tissue (Smith and Nanci, 1995). Tooth development is finalized at the time of its eruption.

The regular distribution patterns of the main components observed in hippopotamid enamel appear to be typical for mammal enamel in general, as indicated by studies of rodents, porcupine, apes or hominoids (Driessens et al., 1990; Verbeeck et al., 1985; Shaw and Yen, 1972; Besic et al., 1969; Johnson, 1972). Thus, although the temporal progression of the different stages of amelogenesis and the ultrastructure of the biominerals differs among species (Smith, 2008), the similar element distribution suggests similar control mechanisms during enamel formation. Secretion and maturation of the enamel mineral control the element distribution in the same way in all taxa, and those processes fractionate the elements to different degrees. The distribution patterns of the minor elements in enamel cannot reflect variations in the diet because these different taxa live in very different habitats and have very different food requirements which render it rather implausible that their diet changes in a similar way during the time of tooth formation.

The functions of an element at the cellular or physiological level involve reactions among minerals and inorganic or organic species in the fluid phase and ultimately are attributable to interactions at the molecular level. However, the way the fluxes of the different ions, in particular those of Na, Mg and Cl, physiologically organized across the

5221

ameloblast layer is not well known. In healthy animal, large concentration variations of CaO and P<sub>2</sub>O<sub>5</sub> in the enamel fluid will not occur at any stage of enamel formation, because these components are stoichiometrically controlled and minor changes in fluid composition alter supersaturating conditions and might trigger the precipitation of different apatite phases (Aoba and Moreno, 1992). The epithelial Ca transport in and out of the mineralizing zone has been reviewed by Hubbard (2000) and Bawden (1989). Calcium is transported through the ameloblast layer either actively through the ameloblast cell (transcellular) or passively through the junctional complexes between ameloblast cells (paracellular). The first transport mode appears to dominate during the secretion phase and provides a controlled low Ca flux which maintains a stable environment for appositional growth and the formation of apatite crystallites in the organic matrix until the final size of the enamel body is reached. During this stage the enamel fluid has much lower Ca, Mg and Na concentrations than the interstitial fluid (Aoba and Moreno, 1987; Moreno and Aoba, 1987). However, during the secretion stage concentration and distribution patterns from the EDJ towards the outer enamel of Na, Mg and Cl vary significantly (Sabel et al., 2009).

At the end of appositional growth and the beginning of the maturation phase the rate of apatite formation increases which suggests a much higher Ca flux during the maturation process, possibly 4 times higher than in the secretory phase (Smith, 1998). At this stage also concentrations of Na, and Mg in enamel appear to be highest (Sabel et al., 2009; Robinson et al., 1995). If maturation begins at the EDJ this explains that enamel near the EDJ has the highest concentrations of these elements. During maturation the ameloblast layer undergo systematic, periodical morphological changes which modulate the ion transport through paracellular and transcellular pathways. Although these processes are not well understood for the different elements, the systematic decrease in Mg, Na in enamel towards the rim implies that there is a decrease of these elements in the maturation fluid with time.

The composition of the enamel fluid filling this space is different from the external fluid and at the various stages of amelogenesis the fluid composition during secretion

5222

and maturation is changing but appears to remain in equilibrium with the crystallizing mineral (Aoba, 1997; Aoba and Moreno, 1992; Hubbard, 2000; Smith, 1998). This implies that the composition of the final, mature enamel apatite represents a mixture of at least 2 components: the apatite produced during secretion plus that added during maturation. This is supported by the distribution pattern of the minor elements Na<sub>2</sub>O, MgO and Cl. There is a common trend for Na<sub>2</sub>O and Cl, but MgO and Na<sub>2</sub>O contents display a fan shape distribution (Fig. 8). The samples near the EDJ show a broad MgO variation which becomes smaller towards the outer enamel rim and lower MgO and Na<sub>2</sub>O concentrations (Fig. 8). In fact, the different populations follow linear trends of rather constant MgO/Na<sub>2</sub>O ratios passing closely the zero point of the diagram. Hence, the MgO/Na<sub>2</sub>O ratio in enamel is well preserved during maturation and the concentration of Na<sub>2</sub>O and MgO follow in a first approximation a binary mixing trend having secretion and maturation fluid as end members. In other words, the flux of Na and Mg ions during maturation through the epithelium is very restricted and is overwhelmed by the flux of major components, such as Ca and P and possibly other anions and result in the crystallization of Mg and Na-poor apatite. The distribution of Cl and Na in all enamel specimens shows one linear trend also resembling binary mixing (Fig. 8). But the higher Cl concentration in the marginal enamel relative to that near the EDJ suggests that Cl increases in the fluid during maturation.

We do not dismiss the existence of at least two different fluid compositions during enamel amelogenesis including the maturation stage. On the other hand, the observed distribution of Mg, Na and Cl in enamel cannot be explained by simple mixing processes but is ultimately the consequence of the mechanism of apatite crystallization during maturation, i.e. the partitioning behavior of these elements between apatite and fluid ( $D_{P/FI}$ ). This is because the enamel body is built during appositional growth and it may only consists of about 14 wt. % of apatite before maturation starts (Smith, 1998). If the enamel formed during secretion is homogeneous in composition and ion fluxes of Na and Mg into the mineralizing compartment are minimized, the subsequent maturation process must include both, formation of new apatite and recrystallization of apatite

5223

precipitated during tooth secretion. In order to explain the observed element distribution, Na, Mg and Cl initially deposited during secretion have to be redistributed within the enamel body.

### Modeling minor element composition of bioapatite

Both, Na and Mg can replace Ca by ion exchange on the crystal surface and initially during the secretion these elements may occur in a labile status, probably adsorbed onto the crystal surfaces (Aoba et al., 1992a, b). However, in order to remove these elements from the fluid they have to be incorporated into the lattice during crystal growth. This effectively means both elements can be accommodated by hydroxyl apatite ( $D_{P/FI} < 1$ ) and they have similar partition coefficients as indicated by the rather constant MgO/Na<sub>2</sub>O ratio in single teeth and individual populations (Figs. 7, 8). In contrast, Cl does not fit well into the hydroxyapatite lattice and becomes highly concentrated in the fluid because the pore space decreases to less than 1%. In this case, assuming a constant and low partition coefficient between apatite and fluid ( $D_{P/FI} < 1$ ), the Cl concentration in apatite must increase with crystallization. The process described above can potentially be quantitatively described by a model assuming equilibrium or fractional crystallization. In this case, equilibrium crystallization presumes that the pores within in the enamel body are connected and concentration gradients in the enamel fluid can be compensated by diffusion. This scenario applies to the beginning of the maturation process in order to allow the redistribution Na, Mg and Cl all over the enamel body. With increasing rate of apatite crystallization, the enamel fluid becomes depleted in Mg and Na but enriched in Cl. As the pore spaces are filled with apatite, the enamel body becomes less permeable. In this case fractional crystallization represents an appropriate model to describe the evolution of fluid and apatite composition until it is completely crystallized.

Figure 8c and d shows models of Rayleigh (fractional) crystallization (Shaw, 1970). for Na<sub>2</sub>O, MgO and Cl variations in enamel from fossil Lake Albert and Lake Kikorongo.

5224

Important parameters such as fluid compositions ( $C_{\text{Fluid}}$ ) and apatite-fluid partition coefficients ( $D_{\text{Ap}/\text{F}}$ ) are not known. However, this example demonstrates in principle that the proposed process of enamel crystallization can explain the variation of the minor elements in enamel. The calculations are summarized in Table 6 using Eqs. (1) and (2) to calculate the changes in concentration in the residual fluid ( $C_{\text{F}}$ ) and solid phase ( $C_{\text{Ap}}$ ), respectively.

In a first step the fluid composition during the secretion process is calculated and this is followed by a simulation of the maturation process by letting the fluid crystallize almost to completeness (99%;  $F=0.01$ , Table 6). The simplest case is to consider the apatite composition near the EDJ as representative for the apatite crystallizing at the end of secretion ( $C_{\text{Apatite}}$  in Table 6; Fig. 8). Assuming  $D_{\text{Ap}/\text{F}}$  of 1.2, 1.2, and 0.5 for  $\text{Na}_2\text{O}$ ,  $\text{MgO}$  and  $\text{Cl}$ , respectively, and that at that time 14 wt. % ( $F=0.86$ ; (Smith, 1998)) of enamel have crystallized. In this case the initial composition of the secretion fluids from Lake Albert and Lake Kikorongo have different  $\text{MgO}$  contents (0.3 wt. %, 0.52 wt. %, respectively; Table 6) but similar  $\text{Cl}$  and  $\text{Na}_2\text{O}$  contents (1.29 wt. %  $\text{Na}_2\text{O}$ , 0.19 wt. %  $\text{Cl}$ ; Table 6). The change in apatite compositions as the fluid crystallizes follows closely the trend observed in our samples and the apatite composition at the end of crystallization is very similar to that observed at the outer rim of the enamel from Lake Albert and Lake Kikorongo (Table 6, Fig. 8). Models of equilibrium crystallization give similar compositional trends, although partition coefficients for  $\text{Na}_2\text{O}$  and  $\text{MgO}$  will be higher ( $\sim 3$ ) and for  $\text{Cl}$  lower (ca. 0.1). One could also consider continued addition of fluid and minor amounts of  $\text{Cl}$ ,  $\text{Mg}$ , and  $\text{Na}$  into the mineralizing compartment using models describing the concomitant crystallization and replenishment (O'Hara and Mathews, 1981). However, at this stage fine-tuning of the models provides no additional information, because important parameters such as fluid composition, ratio of fluid addition to amount of crystallization or the partition coefficient between apatite and fluid are not known.

The model described above implies the maturation process is the dominant process controlling the element distribution in enamel. This is likely, because the main

5225

mass of enamel apatite crystallizes during this phase of tooth evolution (ca. 86 wt. %; Smith (1998)). In detail, the spatial and temporal progression of the maturation within the enamel body is not known. However, the  $\text{MgO}$ ,  $\text{Na}_2\text{O}$  and  $\text{Cl}$  concentration along longitudinal sections near the EDJ are high and rather constant and therefore do not display the typical distribution observed perpendicular to the EDJ (see data for samples Alb10-L1, Mal8-L1, S-276-L1, 5306-L1 in Tables S1 and S3 in the Supplement). This suggests maturation involves the entire tooth but recrystallization and addition of new apatite starts at the EDJ. The proposed recrystallization mechanisms might entail the composition of the enamel near the EDJ to be not representative for the secretion phase. However, because partition coefficients should be similar at all stages of apatite formation,  $\text{MgO}/\text{Na}_2\text{O}$  ratios in the initial apatite should be similar to that observed at the EDJ, and therefore our conclusions regarding the potential of this ratio to distinguish environmental settings are upholding. Future detailed micro analysis with areal resolution distinguishing growth features such as the Retzius lines and maturation fronts may quantitatively resolve the different contributions of secretion and maturation processes on the element distribution in enamel.

## 5 Conclusions

This study is aimed to provide a detailed data base on major and minor element compositions of modern and fossil tooth from hippopotamids in order to (a) identify sedimentary environments, (b) assess chemical distribution patterns induced during enamel formation and (c) to recognize diagenetic overprint.

Secondary alteration controls in particular  $\text{FeO}$ ,  $\text{MnO}$ ,  $\text{SO}_3$  and  $\text{F}$  concentration, which are 2 to 10 times higher in fossil than in modern enamel. The post-mortem addition of these elements appears to be a general phenomenon in fossil bone and tooth materials, irrespective of age and the environmental settings worldwide. Very often the distribution of these elements along enamel profiles perpendicular to the EDJ shows U-shape patterns which are the result of diffusive migration of the surrounding pore solution in the sediment into the tooth (Kohn, 2008; Millard and Hedges, 1996).

5226

The addition of F, FeO, MnO, and SO<sub>3</sub> occurs at constant Ca/P ratios, which implies that secondary Ca-P minerals, such as fluorite or vivianite, do not precipitate during alteration. The concentrations of components such as SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, K<sub>2</sub>O are often low (<0.1 wt. %) in fossil and modern material indicating mechanical introduction and precipitation of silicate minerals has not been relevant. Fossil dentin and cement have lower P<sub>2</sub>O<sub>5</sub> and CaO contents than enamel and even the Ca/P ratio differs from that enamel. However, because of the large variation of the concentrations, including those of Na, Mg and Cl, no systematic difference between fossil and modern dentin and cement can be identified. Systematic studies along dentin growth structure have to be made in order to evaluate whether the observed chemical distributions represent original patterns manifested during dentin and cement formation or whether they have been overprinted by post-mortem processes.

Throughout modern and fossil enamel bodies P<sub>2</sub>O<sub>5</sub> and CaO contents and the CaO/P<sub>2</sub>O<sub>5</sub> ratio are very constant and do not display systematic trends along section perpendicular to the EDJ or along cervix-apex profiles. Linear regression analysis reveals very tight control on the variation of MgO, Na<sub>2</sub>O and Cl ( $R^2=0.64$  to 0.95) notwithstanding large concentration variations across the enamel and regardless of their age and origin. The concentration of MgO and Na<sub>2</sub>O decrease from the EDJ towards the outer rim, whereas Cl displays the opposite variation. As a result, ratios such as MgO/CaO, Na<sub>2</sub>O/CaO or Cl/CaO also systematically decrease or increase towards the enamel rim and are interpreted to reflect changes in the apatite composition rather than changes in mineral density. Similar distribution pattern have been observed in enamel from other mammals like rats, porcupines, apes and humans, which implies that the variation of these elements in enamel apatite is well physiologically controlled and that the fundamental chemical processes of mineralization during amelogenesis follow predominantly the same laws.

During the secretion and maturation phases of amelogenesis the epithelium generates different fluid composition and, in principle, the final composition of enamel apatite is controlled by two fluid compositions. This is supported by co-linear relationships

5227

between MgO, Cl and Na<sub>2</sub>O which can be interpreted to represent binary mixing lines. A plausible explanation for the observed minor element distribution is that both, apatite crystallites precipitating during secretion and apatite crystallizing during maturation equilibrate with a continuously evolving fluid and bioapatite eventually replaces organic matrices and the enamel fluid. This process might be described by equilibrium and fractional crystallization models with continuous replenishment of the mineralizing chamber with maturation fluid.

Despite the opposite behavior of Cl and Na<sub>2</sub>O the variation is co-linear in all specimens and overall differences among enamel from different environments are difficult to resolve. However, MgO concentrations and MgO/Na<sub>2</sub>O ratios in enamel are different among various environments. In particular in river and lake water, which have suffered evaporation (saline environments), enamel has high MgO concentrations and MgO/CaO ratios if compared with their fresh water counterparts. The MgO/Na<sub>2</sub>O ratio in single specimens is rather constant and this makes this ratio a sensitive fingerprint of certain environments.

The proposed process of apatite crystallization during enamel formation, despite representing an oversimplification, poses some questions upon the meaning of other proxies regarding environmental changes and regarding sampling strategies. For example, the oxygen isotope composition is a very sensitive signal of environmental change (temperature, precipitation or aridity), which is effectively transferred into the animal's body fluid. How does the oxygen isotope composition in the different enamel fluids vary during secretion and maturation? If enamel apatite crystallizes in equilibrium with the surrounding, but continuously evolving fluid, how does the oxygen isotope composition at a specific location in the enamel body change with time and which time frame does it represent? Which time frames do we analyze if serial sampling follows transects parallel to or perpendicular to the EDJ? The answers to these questions are important in order to recover meaningful short time-scale information from the incremental growth structures of bioapatite.

5228



*Acknowledgements.* We thank our Ugandan research partners from the Makerere University  
5 in Kampala for their support and discussions on the evolution of the Albertine Rift. Thank you  
to A. Schumann (Kampala) for his help with logistics. K.-H. Becker, M. Maus and A. Budsky  
(University of Mainz) and M. Biegler (MPI Chemie of Mainz) prepared thin sections and mini  
core sample. We thank N. Groschopf for assistance during the microprobe analyses. Funding  
10 by the German Research Foundation (DFG, Research Unit RIFTLINK, project C1) is gratefully  
acknowledged.

## References

- Anderson, P. E., Benton, M. J., Trueman, C. N., Paterson, B. A., and Cuny, G.: Palaeoenviron-  
ments of vertebrates on the southern shore of tethys: The nonmarine early cretaceous of  
Tunisia, *Palaeogeogr. Palaeoclimatol.*, 243, 118–131, 2007.
- 15 Aoba, T.: The effect of fluoride on apatite structure and growth, *Crit. Rev. Oral Biol. M.*, 8,  
136–153, doi:10.1177/10454411970080020301, 1997.
- Aoba, T. and Moreno, E.: The enamel fluid in the early secretory stage of porcine amelogenesis:  
Chemical composition and saturation with respect to enamel mineral, *Calcified Tissue Int.*,  
41, 86–94, doi:10.1007/bf02555250, 1987.
- 20 Aoba, T. and Moreno, E.: Changes in the nature and composition of enamel mineral during  
procine amelogenesis, *Calcified Tissue Int.*, 47, 356–364, doi:10.1007/bf02555887, 1990.
- Aoba, T. and Moreno, E. C.: Changes in the solubility of enamel mineral at various stages of  
procine amelogenesis, *Calcified Tissue Int.*, 50, 266–272, doi:10.1007/bf00296292, 1992.
- Aoba, T., Moreno, E. C., and Shimoda, S.: Competitive adsorption of magnesium and  
25 calcium ions onto synthetic and biological apatites, *Calcified Tissue Int.*, 51, 143–150,  
doi:10.1007/bf00298503, 1992a.
- Aoba, T., Shimoda, S., and Moreno, E. C.: Labile or surface pools of magnesium, sodium,  
and potassium in developing porcine enamel mineral, *J. Dent. Res.*, 71, 1826–1831,  
doi:10.1177/00220345920710111201, 1992b.

5229

- Bahati, G., Pang, Z., Ármannsson, H., Isabirye, E. M., and Kato, V.: Hydrology and reservoir  
characteristics of three geothermal systems in western Uganda, *Geothermics*, 34, 568–591,  
2005.
- Bawden, J. W.: Calcium transport during mineralization, *The Anatomical Record*, 224, 226–  
233, doi:10.1002/ar.1092240212, 1989.
- 5 Besic, F. C., Knowles, C. R., Wiemann, M. R., and Keller, O.: Electron probe microanalysis  
of noncarious enamel and dentin and calcified tissues in mottled teeth, *J. Dent. Res.*, 48,  
131–139, doi:10.1177/00220345690480010501, 1969.
- Beuning, K. R. M., Talbot, M. R., and Kelts, K.: A revised 30,000-year paleoclimatic and paleo-  
10 hydrologic history of Lake Albert, east africa, *Palaeogeogr. Palaeoclimatol.*, 136, 259–279,  
1997.
- Boisserie, J.-R.: The phylogeny and taxonomy of hippopotamidae (mammalia: Artiodactyla): A  
review based on morphology and cladistic analysis, *Zool. J. Linn. Soc.*, 143, 1–26, 2005.
- Boisserie, J. R., Zazzo, A., Merceron, G., Blondel, U., Vignaud, P., Likius, A., Mackaye, H. T.,  
15 and Brunet, M.: Diets of modern and late Miocene hippopotamids: Evidence from carbon  
isotope composition and micro-wear of tooth enamel, *Palaeogeogr. Palaeoclimatol.*, 221,  
153–174, 2005.
- Boskey, A. L.: Mineralization of bones and teeth, *Elements*, 3, 385–391,  
doi:10.2113/gselements.3.6.385, 2007.
- 20 Brachert, T., Brüggmann, G., Mertz, D., Kullmer, O., Schrenk, F., Jacob, D., Ssemmanda, I., and  
Taubald, H.: Stable isotope variation in tooth enamel from Neogene hippopotamids: Monitor  
of meso and global climate and rift dynamics on the Albertine Rift, Uganda, *International J.*  
*Earth Sci.*, 99, 1663–1675, doi:10.1007/s00531-010-0518-1, 2010.
- Buddecke, E.: *Biochemische Grundlagen der Zahnmedizin*, Walter de Gruyter, Berlin, New  
25 York, 192 pp., 1981.
- Cerling, T. E., Harris, J. M., and Passey, B. H.: Diets of east african bovi-  
dae based on stable isotope analysis, *J. Mammal.*, 84, 456–470, doi:10.1644/1545-  
1542(2003)084<0456:DOEABB>2.0.CO;2, 2003.
- Cerling, T. E., Harris, J. M., Hart, J. A., Kaleme, P., Klingel, H., Leakey, M. G., Levin, N. E.,  
30 Lewison, R. L., and Passey, B. H.: Stable isotope ecology of the common hippopotamus, *J.*  
*Zool.*, 276, 204–212, doi:10.1111/j.1469-7998.2008.00450.x, 2008.
- Copeland, S. R., Sponheimer, M., Lee-Thorp, J. A., le Roux, P. J., de Ruiter, D. J., and Richards,  
M. P.: Strontium isotope ratios in fossil teeth from South Africa: Assessing laser ablation mc-

5230

- icp-ms analysis and the extent of diagenesis, *Jo. Archaeol. Sci.*, 37, 1437–1446, 2010.
- Crausbay, S., Russell, J., and Schnurrenberger, D.: A ca. 800-year lithologic record of drought from sub-annually laminated lake sediment, East Java, *J. Paleolimnol.*, 35, 641–659, doi:10.1007/s10933-005-4440-7, 2006.
- 5 Dauphin, Y. and Williams, C. T.: Diagenetic trends of dental tissues, *C. R. Palevol*, 3, 583–590, 2004.
- Dolphin, A. E. and Goodman, A. H.: Maternal diets, nutritional status, and zinc in contemporary mexican infants' teeth: Implications for reconstructing paleodiets, *Am. J. Phys. Anthropol.*, 140, 399–409, 2009.
- 10 Driessens, F. C. M., Goldberg, M., Heijligers, H. J. M., Carreau, J. P., and Verbeeck, R. M. H.: Gradients in the composition of enamel and dentin mineral in rat incisors, *B. Soc. Chim. Belg.*, 99, 861–864, doi:10.1002/bscb.19900991013, 1990.
- Fitch, A., Grauer, A., and Augustine, L.: Lead isotope ratios: Tracking the migration of european-americans to grafton, illinois in the 19th century, *Int. J. Osteoarchaeol.*, n/a–n/a, doi:10.1002/oa.1207, 2010.
- 15 Forbes, M. S., Kohn, M. J., Bestland, E. A., and Wells, R. T.: Late Pleistocene environmental change interpreted from  $\delta^{13}\text{C}$  and  $\delta^{18}\text{O}$  of tooth enamel from the Black Creek Swamp megafauna site, Kangaroo Island, South Australia, *Palaeogeogr. Palaeoclimatol.*, 291, 319–327, 2010.
- 20 Foucault, A. and Stanley, D. J.: Late quaternary palaeoclimatic oscillations in East Africa recorded by heavy minerals in the Nile Delta, *Nature*, 339, 44–46, 1989.
- Fraústo da Silva, J. J. R. and Williams, R. J. P.: *The biological chemistry of the elements : The inorganic chemistry of life*, Oxford Univ. Press, Oxford [u.a.], 2001.
- Fricke, H. C. and O'Neil, J. R.: Inter- and intra-tooth variation in the oxygen isotope composition of mammalian tooth enamel phosphate: Implications for palaeoclimatological and palaeobiological research, *Palaeogeogr. Palaeoclimatol.*, 126, 91–99, 1996.
- 25 Fricke, H. C., Clyde, W. C., O'Neill, J. R., and Gingerich, P. D.: Evidence for rapid climate change in north america during the latest Paleocene thermal maximum: Oxygen isotope compositions of biogenic phosphate from the Bighorn Basin (Wyoming), *Earth Planet. Sci. Lett.*, 160, 193–208, 1998.
- 30 Glimcher, M. J.: Bone: Nature of the calcium phosphate crystals and cellular, structural, and physical chemical mechanisms in their formation, *Rev. Mineral. Geochem.*, 64, 223–282, doi:10.2138/rmg.2006.64.8, 2006.

5231

- Glimcher, M. J., Cohen-Solal, L., Kossiva, D., and de Ricqles, A.: Biochemical analyses of fossil enamel and dentin, *Paleobiology*, 16, 219–232, 1990.
- Grandstaff, D. E. and Terry Jr, D. O.: Rare earth element composition of Paleogene vertebrate fossils from toadstool geologic park, Nebraska, USA, *Appl. Geochem.*, 24, 733–745, 2009.
- 5 Hammer, U. T.: *Saline lake ecosystems of the world*, Monographiae biologicae, 59, Springer, 632 pp., 1986.
- Hubbard, M. J.: Calcium transport across the dental enamel epithelium, *Crit. Rev. Oral Biol. M.*, 11, 437–466, doi:10.1177/10454411000110040401, 2000.
- 10 Jacques, L., Ogle, N., Moussa, I., Kalin, R., Vignaud, P., Brunet, M., and Bocherens, H.: Implications of diagenesis for the isotopic analysis of upper Miocene large mammalian herbivore tooth enamel from Chad, *Palaeogeogr. Palaeoclimatol.*, 266, 200–210, 2008.
- Johnson, A. R.: Strontium, calcium, magnesium, and phosphorus content of rat incisors as determined by electron microprobe analysis, *J. Dent. Res.*, 51, 115–121, doi:10.1177/00220345720510010301, 1972.
- 15 Kempe, S.: Damming the Nile, *Mitt. Geol.-Paläont. Ist. Univ. Hamburg*, 74, 81–114, 1993.
- Koch, P. L.: Isotopic reconstruction of past continental environments, *Annu. Rev. Earth Pl. Sc.*, 26, 573–613, doi:10.1146/annurev.earth.26.1.573, 1998.
- Koch, P. L., Diffenbaugh, N. S., and Hoppe, K. A.: The effects of late quaternary climate and pCO<sub>2</sub> change on C<sub>4</sub> plant abundance in the South-Central United States, *Palaeogeogr. Palaeoclimatol.*, 207, 331–357, 2004.
- 20 Kohn, M. J.: Predicting animal  $\delta^{18}\text{O}$ : Accounting for diet and physiological adaptation, *Geochim. Cosmochim. Ac.*, 60, 4811–4829, 1996.
- Kohn, M. J.: Models of diffusion-limited uptake of trace elements in fossils and rates of fossilization, *Geochim. Cosmochim. Ac.*, 72, 3758–3770, 2008.
- 25 Kohn, M. J. and Cerling, T. E.: Stable isotope compositions of biological apatite, in: *Phosphates – geochemical, geobiological, and materials importance*, edited by: Kohn, M. J., Rakovan, J., and Hughes, J. M., Reviews in mineralogy and geochemistry, Mineralogical Society of America, Washington, DC, 455–488, 2002.
- Kohn, M. J., Schoeninger, M. J., and Valley, J. W.: Herbivore tooth oxygen isotope compositions: Effects of diet and physiology, *Geochim. Cosmochim. Ac.*, 60, 3889–3896, 1996.
- 30 Kohn, M. J., Schoeninger, M. J., and Barker, W. W.: Altered states: Effects of diagenesis on fossil tooth chemistry, *Geochim. Cosmochim. Ac.*, 63, 2737–2747, 1999.
- Larsson, P. A., Howell, D. S., Pita, J. C., and Blanco, L. N.: Aspiration and characterization

5232

- of predentin fluid in developing rat teeth by means of a micropuncture and micro-analytical technique, *J. Dent. Res.*, 67, 870–875, doi:10.1177/00220345880670051501, 1988.
- Lundgren, T., Persson, L. G., Engström, E. U., Chabala, J., Levi-Setti, R., and Norén, J. G.: A secondary ion mass spectroscopic study of the elemental composition pattern in rat incisor dental enamel during different stages of ameloblast differentiation, *Arch. Oral Biol.*, 43, 841–848, 1998.
- MacFadden, B. J., Solounias, N., and Cerling, T. E.: Ancient diets, ecology, and extinction of 5-million-year-old horses from Florida, *Science*, 283, 824–827, doi:10.1126/science.283.5403.824, 1999.
- Margolis, H. C., Beniash, E., and Fowler, C. E.: Role of macromolecular assembly of enamel matrix proteins in enamel formation, *J. Dent. Res.*, 85, 775–793, doi:10.1177/154405910608500902, 2006.
- Millard, A. R. and Hedges, R. E. M.: A diffusion-adsorption model of uranium uptake by archaeological bone, *Geochim. Cosmochim. Ac.*, 60, 2139–2152, 1996.
- Moreno, E. C. and Aoba, T.: Calcium binding in enamel fluid and driving force for enamel mineralization in the secretory stage of amelogenesis, *Adv. Dent. Res.*, 1, 245–251, 1987.
- Norén, J. G., Lodding, A., Odellius, H., and Linde, A.: Secondary ion mass spectrometry of human deciduous enamel, *Caries Res.*, 17, 496–502, 1983.
- O'Hara, M. J. and Mathews, R. E.: Geochemical evolution in an advancing, periodically replenished, periodically tapped, continuously fractionated magma chamber, *J. Geol. Soc.*, 138, 237–277, doi:10.1144/gsjgs.138.3.0237, 1981.
- Palmer, M. R. and Edmond, J. M.: The strontium isotope budget of the modern ocean, *Earth Planet. Sci. Lett.*, 92, 11–26, 1989.
- Pasteris, J. D., Wopenka, B., and Valsami-Jones, E.: Bone and tooth mineralization: Why apatite?, *ELEMENTS*, 4, 97–104, doi:10.2113/gselements.4.2.97, 2008.
- Patrick, D., Martin, J. E., Parris, D. C., and Grandstaff, D. E.: Paleoenvironmental interpretations of rare earth element signatures in mosasaurs (reptilia) from the Upper Cretaceous Pierre Shale, Central South Dakota, USA, *Palaeogeogr. Palaeoclimatol.*, 212, 277–294, 2004.
- Pickford, M., Senut, B., and Hadoto, D.: Geology and palaeontology of the Albertine Rift Valley, Uganda-Zaire, Publication Occasionelle, Centre International pour la Formation et les Echanges Geologiques, Orléans Cedex 2, 24, 1–190, 1993.
- Pike, A. W. G., Hedges, R. E. M., and Van calsteren, P.: U-series dating of bone using the diffusion-adsorption model, *Geochim. Cosmochim. Ac.*, 66, 4273–4286, 2002.

5233

- Revel, M., Ducassou, E., Grousset, F. E., Bernasconi, S. M., Migeon, S., Revillon, S., Mascle, J., Murat, A., Zaragosi, S., and Bosch, D.: 100,000 years of African monsoon variability recorded in sediments of the Nile margin, *Quaternary Sci. Rev.*, 29, 1342–1362, 2010.
- Robinson, C., Kirkham, J., Brookes, S. J., Bonass, W. A., and Shore, R. C.: The chemistry of enamel development, *Int. J. Dev. Biol.*, 39, 145–115, 1995.
- Sabel, N., Klinberg, G., Nietzsche, S., Robertson, A., Odellius, H., and Noren, J.: Analysis of some elements in primary enamel during postnatal mineralization, *Swed. Dent. J.*, 33, 85–95, 2009.
- Safont, S., Malgosa, A., Subirà, M., and Gibert, J.: Can trace elements in fossils provide information about palaeodiet?, *Int. J. Osteoarchaeol.*, 8, 23–37, doi:10.1002/(sici)1099-1212(199801/02)8:1<23::aid-oea403>3.0.co;2-r, 1998.
- Salama, R. B. and Selley, R. C.: Chapter 6 rift basins of the Sudan, in: *Sedimentary basins of the world*, Elsevier, 105–149, 1997.
- Scrivner, A. E., Vance, D., and Rohling, E. J.: New neodymium isotope data quantify Nile involvement in mediterranean anoxic episodes, *Geology*, 32, 565–568, doi:10.1130/g20419.1, 2004.
- Senut, B. and Pickford, M.: Geology and palaeobiology of the Albertine Rift Valley, Uganda-Zaire, Volume ii: Palaeobiology, in: *Publication occasionelle – centre international pour la formation et les echanges geologiques CIFEG Orléans*, 1–423, 1994.
- Shaw, D. M.: Trace element fractionation during anatexis, *Geochim. Cosmochim. Ac.*, 34, 237–243, 1970.
- Shaw, J. H. and Yen, P. K.-J.: Sodium, potassium, and magnesium concentrations in the enamel and dentin of human and rhesus monkey teeth, *J. Dent. Res.*, 51, 95–101, doi:10.1177/00220345720510013701, 1972.
- Sillen, A. and Kavanagh, M.: Strontium and paleodietary research: A review, *Am. J. Phys. Anthropol.*, 25, 67–90, 1982.
- Sillen, A.: Elemental and isotopic analyses of mammalian fauna from Southern Africa and their implications for paleodietary research, *Am. J. Phys. Anthropol.*, 76, 49–60, 1988.
- Simmer, J. P. and Fincham, A. G.: Molecular mechanisms of dental enamel formation, *Crit. Rev. Oral Biol. M.*, 6, 84–108, doi:10.1177/10454411950060020701, 1995.
- Simmer, J. P., Papagerakis, P., Smith, C. E., Fisher, D. C., Rountrey, A. N., Zheng, L., and Hu, J. C.-C.: Regulation of dental enamel shape and hardness, *J. Dent. Res.*, 89, 1024–1038, doi:10.1177/0022034510375829, 2010.

5234

- Simonetti, A., Buzon, M., and Creaser, R.: In-situ elemental and Sr isotope investigations of human tooth enamel by laser ablation -(MC)-ICP-MS: Successes and pitfalls, *Archaeometry*, 50, 371–385, 2008.
- Smith, C. E.: Cellular and chemical events during enamel maturation, *Crit. Rev. Oral Biol. M.*, 9, 128–161, doi:10.1177/10454411980090020101, 1998.
- 5 Smith, C. E. and Nanci, A.: Overview of morphological changes in enamel organ cells associated with major events in amelogenesis., *Int. J. Dev. Biol.*, 39, 153–161, 1995.
- Smith, T. M.: Incremental dental development: Methods and applications in hominoid evolutionary studies, *J. Hum. Evol.*, 54, 205–224, 2008.
- 10 Sponheimer, M. and Lee-Thorp, J. A.: Enamel diagenesis at South African australopith sites: Implications for paleoecological reconstruction with trace elements, *Geochim. Cosmochim. Ac.*, 70, 1644–1654, 2006.
- Sponheimer, M., Lee-Thorp, J. A., DeRuiter, D. J., Smith, J. M., van der Merwe, N. J., Reed, K., Grant, C. C., Ayliffe, L. K., Robinson, T. F., Heidelberg, C., and Marcus, W.: Diets of Southern African bovidae: Stable isotope evidence, *J. Mammal.*, 84, 471–479, doi:10.1644/1545-1542(2003)084<0471:DOSABS>2.0.CO;2, 2003.
- 15 Steinfort, J., Driessens, F. C. M., Heijligers, H. J. M., and Beertsen, W.: The distribution of magnesium in developing rat incisor dentin, *J. Dent. Res.*, 70, 187–191, doi:10.1177/00220345910700030601, 1991.
- 20 Talbot, M. R., Williams, M. A. J., and Adamson, D. A.: Strontium isotope evidence for late Pleistocene reestablishment of an integrated Nile drainage network, *Geology*, 28, 343–346, doi:10.1130/0091-7613(2000)28<343:sieflp>2.0.co;2, 2000.
- Trueman, C. N., Behrensmeyer, A. K., Potts, R., and Tuross, N.: High-resolution records of location and stratigraphic provenance from the rare earth element composition of fossil bones, *Geochim. Cosmochim. Ac.*, 70, 4343–4355, 2006.
- 25 Tütken, T., Vennemann, T. W., and Pfretzschner, H. U.: Early diagenesis of bone and tooth apatite in fluvial and marine settings: Constraints from combined oxygen isotope, nitrogen and REE analysis, *Palaeogeogr. Palaeoclimatol.*, 266, 254–268, 2008.
- Van Damme, D. and Pickford, M.: The Late Cenozoic thiaridae (mollusca, gastropoda, cerithioidea) of the Albertine Rift Valley (Uganda-Congo) and their bearing on the origin and evolution of the Tanganyikan thalassoid malacofauna, *Hydrobiologia*, 498, 1–83, doi:10.1023/a:1026298512117, 2003.
- 30 Verbeek, R. M. H., Driessens, F. C. M., Borggreven, J. M. P. M., and Wöltgens, J. H. M.:

5235

- Concentration gradients of some minor components in human tooth enamel, *B. Soc. Chim. Belg.*, 94, 237–243, doi:10.1002/bscb.19850940402, 1985.
- Weatherell, J. A., Robinson, C., and Hallsworth, A. S.: Variations in the chemical composition of human enamel, *J. Dent. Res.*, 53, 180–192, doi:10.1177/00220345740530020501, 1974.
- 5 Williams, M. A. J.: Late Pleistocene and Holocene environments in the Nile Basin, *Global Planet. Change*, 69, 1–15, 2009.
- Wopenka, B. and Pasteris, J. D.: A mineralogical perspective on the apatite in bone, *Mater. Sci. Eng. C*, 25, 131–143, 2005.
- Wright, J., Schrader, H., and Holser, W. T.: Paleoredox variations in ancient oceans recorded by rare earth elements in fossil apatite, *Geochim. Cosmochim. Ac.*, 51, 631–644, 1987.
- 10 Yu, Z., Ito, E., and Engstrom, D. R.: Water isotopic and hydrochemical evolution of a lake chain in the Northern Great Plains and its paleoclimatic implications, *J. Paleolimnol.*, 28, 207–217, doi:10.1023/a:1021651824885, 2002.
- Zazzo, A., Balasse, M., and Patterson, W. P.: High-resolution  $\delta^{13}\text{C}$  intratooth profiles in bovine enamel: Implications for mineralization pattern and isotopic attenuation, *Geochim. Cosmochim. Ac.*, 69, 3631–3642, 2005.
- 15

5236

**Table 1.** Average chemical composition of enamel from modern and fossil hippopotamid teeth.

Enamel	Lake Albert Modern	Lake Malawi Modern	Zoo S-276	Nile Blue	Nile White	Nile Upper	Lake Kikorongo	Lake Albert Fossil
Specimen/ Analyses	1/120	1/106	1/162	1/24	1/29	2/50	1/156	18/964
P <sub>2</sub> O <sub>5</sub> wt. %	42.81	40.96	42.06	42.57	42.35	42.57	41.30	41.38
SiO <sub>2</sub>	< 0.04	0.56	0.14	0.07	0.05	0.05	0.10	0.13
FeO	0.06	0.06	0.07	0.06	0.05	0.05	0.08	0.66
CaO	53.38	54.69	54.31	53.90	53.89	54.03	54.40	54.16
MgO	0.46	0.49	0.35	0.27	0.43	0.25	0.43	0.22
MnO	< 0.05	< 0.05	0.08	< 0.05	0.06	0.07	0.10	0.26
SrO	0.08	0.08	0.07	0.08	0.08	0.07	0.34	0.08
Na <sub>2</sub> O	1.16	1.19	1.01	1.09	1.18	1.03	1.04	0.94
F	0.16	0.23	0.14	0.11	0.11	0.11	0.26	0.40
Cl	0.31	0.34	0.40	0.46	0.40	0.43	0.34	0.35
SO <sub>3</sub>	0.06	0.10	0.09	0.06	0.06	0.05	0.09	0.17
K <sub>2</sub> O	0.04	0.03	0.03	0.05	0.05	0.05	0.04	0.04
Al <sub>2</sub> O <sub>3</sub>	0.04	0.04	0.09	0.18	0.05	0.05	0.34	0.07
H <sub>2</sub> O	1.59	1.54	1.63	1.62	1.63	1.62	1.52	1.41
Total	92.89	91.49	92.53	93.30	93.21	93.10	92.01	92.18
CaO/ P <sub>2</sub> O <sub>5</sub>	1.25	1.33	1.29	1.27	1.27	1.27	1.32	1.31
MgO/ P <sub>2</sub> O <sub>5</sub>	0.0108	0.0111	0.0083	0.0062	0.0101	0.0058	0.0105	0.0053
FeO/ P <sub>2</sub> O <sub>5</sub>	0.0013	0.0015	0.0018	–	0.0013	0.0012	0.0017	0.0288
Na <sub>2</sub> O/ P <sub>2</sub> O <sub>5</sub>	0.0271	0.0288	0.0241	0.0257	0.0278	0.0243	0.0251	0.0226
F/ P <sub>2</sub> O <sub>5</sub>	0.0037	0.0055	–	–	–	0.0015	0.0062	0.0111
Cl/ P <sub>2</sub> O <sub>5</sub>	0.0072	0.0085	0.0095	0.0108	0.0094	0.0102	0.0083	0.0082
SO <sub>3</sub> / P <sub>2</sub> O <sub>5</sub>	0.0013	0.0022	0.0021	0.0014	0.0014	0.0012	0.0020	0.0038
MgO/Na <sub>2</sub> O	0.41	0.39	0.35	0.25	0.36	0.24	0.42	0.24

5237

**Table 2.** Average chemical composition of dentin from modern and fossil hippopotamid teeth

Dentin	Lake Albert Modern	Lake Malawi Modern	Nile River	Zoo S-276	Lake Kikorongo	Lake Albert Fossil
Specimen/ Analyses	1/25	1/20	1/6	1/13	1/25	18/136
P <sub>2</sub> O <sub>5</sub> wt. %	42.52	38.05	38.65	44.45	36.64	38.29
SiO <sub>2</sub>	< .04	2.99	< .04	0.13	0.13	0.43
FeO	< 0.02	0.05	0.06	0.08	0.21	3.55
CaO	50.55	52.58	54.87	49.47	57.08	52.75
MgO	3.41	2.53	3.00	2.38	0.93	0.19
MnO	< 0.05	< 0.05	< 0.05	< 0.05	0.13	0.73
SrO	0.09	0.09	0.03	0.08	0.72	0.20
Na <sub>2</sub> O	1.12	1.36	1.14	1.11	1.00	0.55
F	0.31	0.33	0.14	0.20	1.27	2.39
Cl	0.06	0.09	0.11	0.22	0.09	0.03
SO <sub>3</sub>	0.28	0.34	0.36	0.47	0.25	0.32
K <sub>2</sub> O	0.04	0.04	0.04	0.04	0.05	0.03
Al <sub>2</sub> O <sub>3</sub>	< 0.03	0.05	0.15	0.10	0.59	0.12
H <sub>2</sub> O	1.64	1.59	1.68	1.72	1.16	0.53
Total	70.80	77.76	77.49	75.27	82.08	88.61
CaO/ P <sub>2</sub> O <sub>5</sub>	1.19	1.38	1.42	1.11	1.62	1.38
MgO/ P <sub>2</sub> O <sub>5</sub>	0.0801	0.0667	0.0777	0.053	0.0282	0.0051
FeO/ P <sub>2</sub> O <sub>5</sub>	–	0.0014	–	0.0019	0.0037	0.0906
Na <sub>2</sub> O/ P <sub>2</sub> O <sub>5</sub>	0.0262	0.0358	0.0295	0.0249	0.0274	0.0147
F/ P <sub>2</sub> O <sub>5</sub>	0.0073	0.0086	0.0047	0.0066	0.0352	0.0612
Cl/ P <sub>2</sub> O <sub>5</sub>	0.0014	0.0023	0.0030	0.0049	0.0022	0.0009
SO <sub>3</sub> / P <sub>2</sub> O <sub>5</sub>	0.0066	0.0090	0.0101	0.0106	0.0070	0.0082
MgO/Na <sub>2</sub> O	3.05	1.85	2.64	2.12	1.11	0.36

5238

**Table 3.** Chemical composition of cement from hippopotamid molar teeth

Cement	Zoo S-276	Lake Malawi	Lake Albert Modern	Lake Albert Fossil	Lake Kikorongo
Specimen/ Analyses	1/16	1/5	1/3	1/15	1/6
P <sub>2</sub> O <sub>5</sub> (wt.%)	42.65	37.16	38.90	39.19	39.02
SiO <sub>2</sub>	0.54	2.00	< 0.04	0.66	1.24
FeO	0.11	0.06	< 0.02	4.09	0.51
CaO	51.63	55.65	56.42	52.11	54.16
MgO	1.17	1.28	1.48	0.20	0.76
MnO	< 0.05	< 0.05	< 0.05	0.59	0.08
SrO	0.10	0.08	0.11	0.15	0.42
Na <sub>2</sub> O	1.02	1.05	0.80	0.45	1.02
F	0.20	0.66	0.34	2.97	0.67
Cl	0.19	0.09	0.03	0.02	0.50
SO <sub>3</sub>	0.76	0.49	0.34	0.48	0.15
K <sub>2</sub> O	0.04	0.05	< 0.02	0.07	0.14
Al <sub>2</sub> O <sub>3</sub>	0.27	0.05	< 0.03	0.74	0.06
H <sub>2</sub> O	1.69	1.46	1.58	0.42	1.31
Total	69.14	74.53	71.72	90.39	90.86
CaO/P <sub>2</sub> O <sub>5</sub>	1.21	1.50	1.45	1.39	1.39
MgO/ P <sub>2</sub> O <sub>5</sub>	0.0274	0.0344	0.0379	0.0055	0.0198
FeO/ P <sub>2</sub> O <sub>5</sub>	0.0025	0.0016	–	0.1109	0.0131
Na <sub>2</sub> O/ P <sub>2</sub> O <sub>5</sub>	0.0240	0.0282	0.0206	0.0121	0.0262
F/ P <sub>2</sub> O <sub>5</sub>	0.0073	0.0177	0.0087	0.0792	0.0175
Cl/ P <sub>2</sub> O <sub>5</sub>	0.0044	0.0024	0.0008	0.0006	0.0127
SO <sub>3</sub> / P <sub>2</sub> O <sub>5</sub>	0.0179	0.0131	0.0088	0.0127	0.0040
MgO/Na <sub>2</sub> O	1.17	1.23	1.85	0.45	0.74

5239

**Table 4.** Regression analysis of the element distribution in enamel along profiles from the outer enamel rim ( $x = 0$ ) towards the enamel-dentin junction ( $x = 1$ ).

Sample	Element	R <sup>2</sup>	m	x = 1 EDJ	x = 0 Outer Enamel	% x1-x0
Modern Ave	Total	0.51	-2.40	91.33 (1.00)	93.73 (1.04)	-2.63
Fossil L-Albert Ave		0.63	-2.59	91.42 (1.35)	94.01 (1.18)	-2.84
Kikorongo Ave		0.54	-2.04	92.56 (0.77)	94.60 (0.39)	-2.21
Modern Ave	CaO	0.15	-0.11	54.06 (0.78)	54.17 (0.46)	-0.22
Fossil L-Albert Ave		0.17	-0.01	54.05 (0.58)	54.06 (0.49)	-0.03
Kikorongo Ave		0.32	-0.54	54.01 (0.24)	54.55 (0.11)	-1.01
Modern Ave	P <sub>2</sub> O <sub>5</sub>	0.16	-0.31	41.97 (0.93)	42.28 (0.59)	-0.76
Fossil L-Albert Ave		0.16	-0.08	41.45 (0.35)	41.52 (0.37)	-0.19
Kikorongo Ave		0.11	-0.10	41.47 (0.47)	41.58 (0.17)	-0.25
Modern Ave	MgO	0.58	0.17	0.44 (0.11)	0.27 (0.07)	37.9
Fossil L-Albert Ave		0.44	0.09	0.26 (0.03)	0.18 (0.03)	32.1
Kikorongo Ave		0.75	0.23	0.56 (0.08)	0.33 (0.03)	40.6
Modern Ave	SrO	0.14	0.02	0.07 (0.02)	0.05 (0.02)	30.0
Fossil L-Albert Ave		0.20	0.04	0.08 (0.03)	0.04 (0.02)	20.3
Kikorongo Ave		0.76	0.21	0.49 (0.11)	0.27 (0.03)	42.7
Modern Ave	Na <sub>2</sub> O	0.84	0.56	1.35 (0.12)	0.79 (0.08)	40.9
Fossil L-Albert Ave		0.85	0.53	1.21 (0.10)	0.68 (0.10)	43.8
Kikorongo Ave		0.90	0.60	1.39 (0.14)	0.79 (0.02)	42.6
Modern Ave	Cl	0.95	-0.41	0.20 (0.05)	0.61 (0.05)	-223
Fossil L-Albert Ave		0.93	-0.38	0.16 (0.05)	0.54 (0.06)	-293
Kikorongo Ave		0.97	-0.40	0.13 (0.01)	0.53 (0.03)	-308
Modern Ave	F	0.13	0.00	0.08 (0.11)	0.20 (0.02)	-3.5
Fossil L-Albert Ave		0.31	0.00	0.37 (0.20)	0.37 (0.25)	-20.1
Kikorongo Ave		0.31	0.12	0.32 (0.10)	0.20 (0.06)	38.2
Modern Ave	SO <sub>3</sub>	0.11	-0.03	0.03 (0.04)	0.06 (0.04)	238
Fossil L-Albert Ave		0.24	-0.02	0.09 (0.12)	0.18 (0.16)	-189
Kikorongo Ave		0.10	0.01	0.08 (0.01)	0.07 (0.02)	10.3
Modern Ave	FeO	0.03	0.00	0.01 (0.01)	0.01 (0.01)	-130
Fossil L-Albert Ave		0.39	-0.16	0.55 (0.33)	0.71 (0.45)	99
Kikorongo Ave		0.09	-0.02	0.02 (0.01)	0.04 (0.05)	-166

Values in parentheses: 1 standard deviation;  $m$ : slope of regression line;  $R^2$ : coefficient of determination

5240

**Table 5.** One Factor analysis of variance (ANOVA) of Enamel and Scheffé tests distinguishing modern, Lake Kikorongo and fossil Lake Albert populations.

	ANOVA <i>p</i> -value	Scheffé test <i>p</i> -value	Scheffé test <i>p</i> -value	Scheffé test <i>p</i> -value
$\alpha = 0.05$	3 populations: – Modern – Kikorongo – Fossil L.Albert	2 populations: – Modern – Fossil L.Albert	2 populations: – Modern – Kikorongo	2 populations: – Fossil L.Albert – Kikorongo
Total	0.2982			
CaO	0.9600			
P <sub>2</sub> O <sub>5</sub>	0.0571			
MgO	<b>&lt;0.0001</b>	<b>0.0002</b>	<b>0.0081</b>	<b>&lt;0.0001</b>
Sr	<b>&lt;0.0001</b>	<b>0.0285</b>	<b>0.0000</b>	<b>&lt;0.0001</b>
Na <sub>2</sub> O	<b>0.0007</b>	<b>0.0003</b>	0.2895	<b>0.0045</b>
Cl	<b>0.0451</b>	<b>0.0186</b>	<b>0.0202</b>	0.1696
F	<b>0.0001</b>	<b>&lt;0.0001</b>	<b>0.0184</b>	0.2950
SO <sub>3</sub>	0.2547			
FeO	<b>&lt;0.0001</b>	<b>&lt;0.0001</b>	0.4893	<b>0.0012</b>

$p < 0.05$  (bold numbers) implies the population means are different, rejecting the H<sub>0</sub> hypothesis of equal variances.

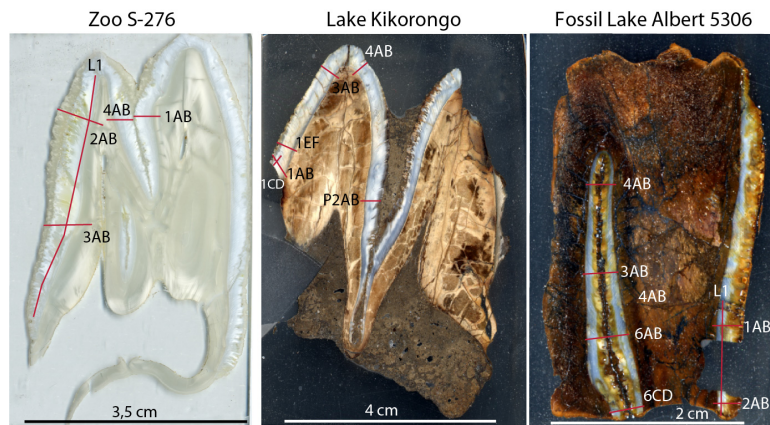
5241

**Table 6.** Rayleigh crystallization model for enamel formation

	F	Na <sub>2</sub> O	MgO	Cl	Na <sub>2</sub> O	MgO	Cl
	Lake Kikorongo				Fossil Lake Albert		
Calculation of fluid composition during secretion							
D <sub>Ap/Fl</sub>		1.2	1.2	0.5	1.2	1.2	0.5
C <sub>Apatite</sub> wt. %		1.5	0.6	0.1	1.5	0.35	0.1
C <sub>fluid</sub> wt. %	0.86	1.29	0.52	0.19	1.29	0.30	0.19
Calculation of apatite composition during maturation							
C <sub>Apatite</sub> wt. %	0.90	1.51	0.61	0.10	1.51	0.35	0.10
	0.80	1.48	0.59	0.10	1.35	0.31	0.13
	0.70	1.44	0.58	0.11	1.22	0.28	0.17
	0.60	1.40	0.56	0.12	0.98	0.23	0.29
	0.50	1.35	0.54	0.13	0.85	0.20	0.41
	0.40	1.29	0.51	0.15	0.62	0.14	0.93
	0.30	1.22	0.49	0.17	0.54	0.13	1.31
	0.20	1.12	0.45	0.21	1.12	0.26	0.21
	0.10	0.98	0.39	0.29	0.98	0.23	0.29
	0.05	0.85	0.34	0.41	0.85	0.20	0.41
	0.01	0.62	0.25	0.93	0.62	0.14	0.93

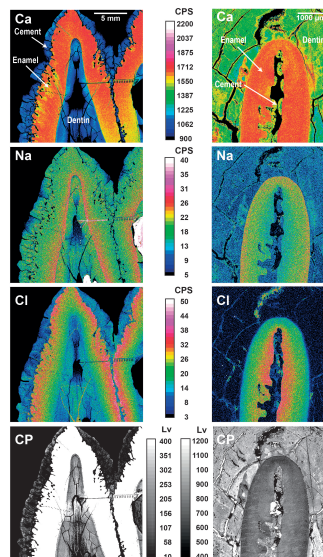
F: fraction of fluid remaining D: partition coefficient Apatite/Fluid  
C: concentration (wt. %) of fluid or apatite.

5242



**Fig. 1.** Thin sections of selected samples of molar hippopotamid teeth ranging in age from recent (S-276), 10 ka (Lake Kikorongo) to 2.3 Ma (5306) indicate variable degrees of diagenetic overprint. Profiles of major and minor element analyses are indicated.

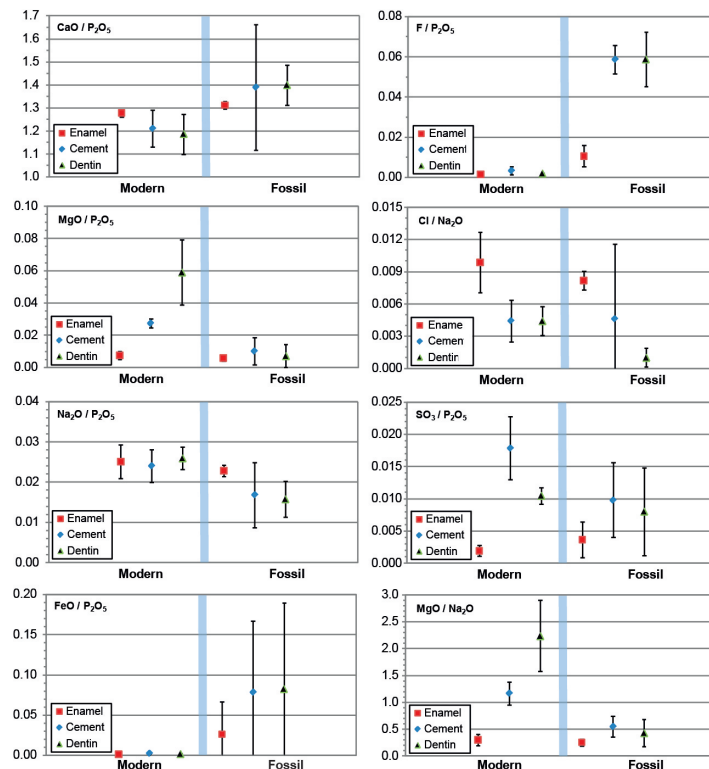
5243



**Fig. 2.** Element maps for Ca, Na, Cl and the backscattered images (CP) of modern (sample S-276, left) and fossil (sample 5306, right) hippopotamid teeth. Note the significantly higher Ca intensities in enamel compared to that in dentin and cement, which reflects lower porosity and higher Ca concentrations in enamel. Variation of Ca along an apex-cervix profile reflects differences in mineral density. The zonal distribution of Cl and Na in enamel reflects concentration variation. Note the contrasting brightness levels (CP) in modern and fossil enamel and dentine. Modern dentine has a low average atomic number (low Z) compared to enamel. Intensities are reversed in fossil materials.

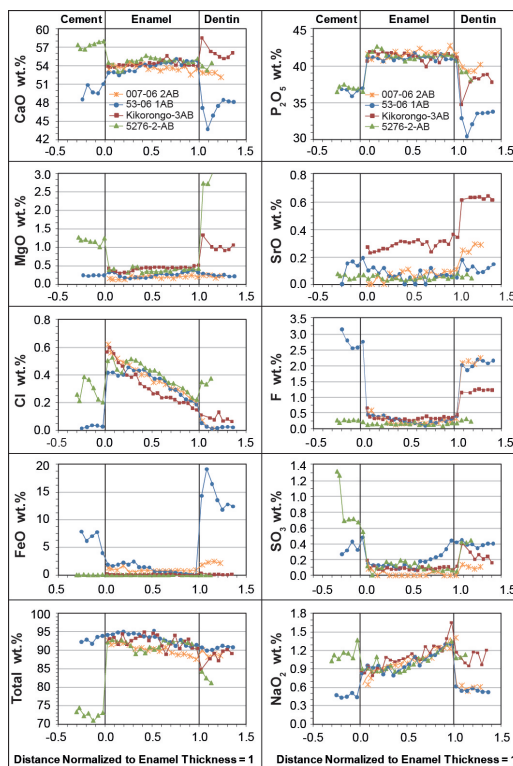
5244





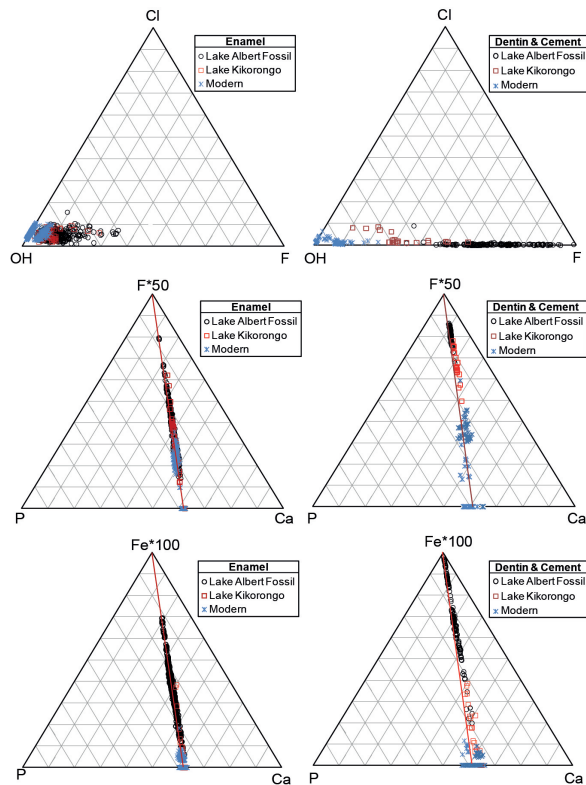
**Fig. 3.** Comparison of the chemical composition of modern and fossil tooth materials from hippopotamids.

5245



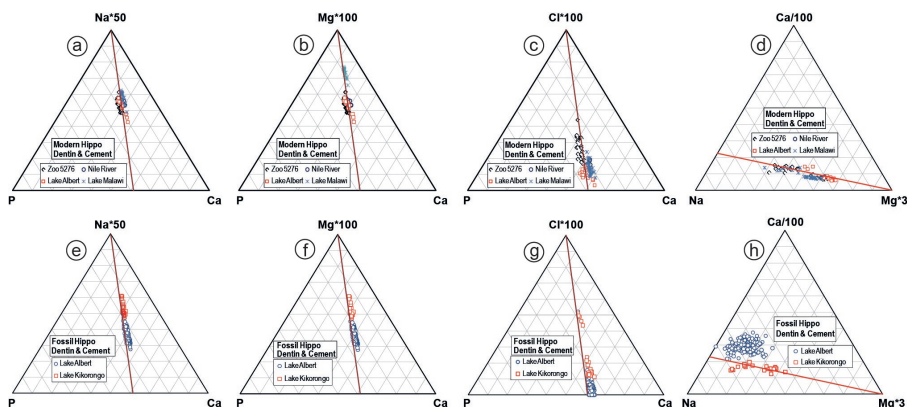
**Fig. 4.** Chemical variations along selected tooth profiles from the outside margin through cementum, enamel into dentin. The distance is normalized to the length of the enamel = 1. Location of the profiles is shown in Fig. 1.

5246



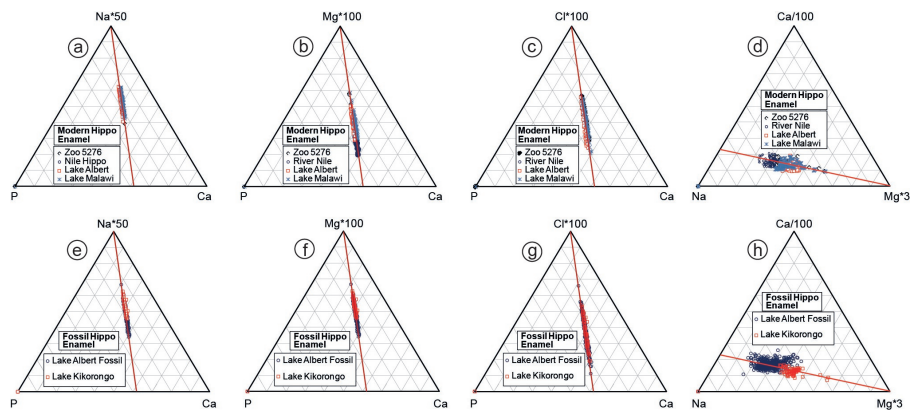
**Fig. 5.** Chemical components (atomic proportions) in enamel, dentin and cement of modern and fossil Hippopotamid teeth indicating the addition of FeO and F during post-mortem alteration processes. Red line defines the trend of modern enamel having a rather constant Ca/P.

5247



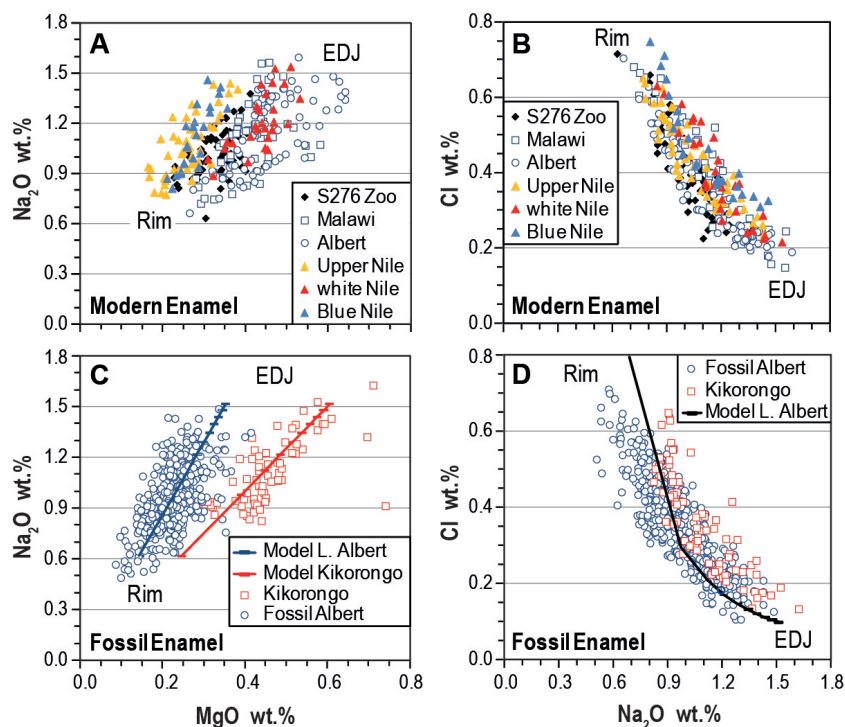
**Fig. 6.** Triangular diagram of chemical components (atomic proportions) in dentin and cement of modern and fossil hippopotamid teeth. Dashed line defines trend of constant P/Ca ratio observed in modern enamel.

5248



**Fig. 7.** Triangular diagram of chemical components (atomic proportions) in enamel of modern and fossil hippopotamid teeth. Red line defines trend of constant P/Ca ratio observed in modern enamel. Modern and fossil enamel have identical P/Ca and display similar variations in Na, Mg, Cl, but variable Mg/Ca and Na/Mg.

5249



**Fig. 8.** Variation of MgO, Na<sub>2</sub>O, and Cl in modern and fossil enamel from Hippopotamid teeth. Note the coherent variation of Cl and Na<sub>2</sub>O in fossil and modern enamel but distinct MgO contents in enamel from different environments. Blue and red lines represent model calculation assuming fractional crystallization of apatite. For more details see text and Table 6.

5250