

Supporting Information for ”Comparison of the SAWNUC model with CLOUD measurements of sulphuric acid-water nucleation”

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1. Figures S1 to S4

Text S1. The SAWNUC model was originally described in *Lovejoy et al.* [2004]. Table S4 summarises the current state of SAWNUC and provides the literature sources containing further details. The reaction schemes, forward reaction rate constants and evaporation rate constants presented there are still used, with the following exceptions:

- The evaporation rates of neutral particles was adjusted in *Kazil and Lovejoy* [2007].
- Instead of calculating a diurnal cycle of H_2SO_4 the measured sulphuric acid concentrations are used.
- The ion-ion recombination rate constant for small ions from *Brasseur and Chatel* [1983] was used.
- A wall loss term was added as first order loss term

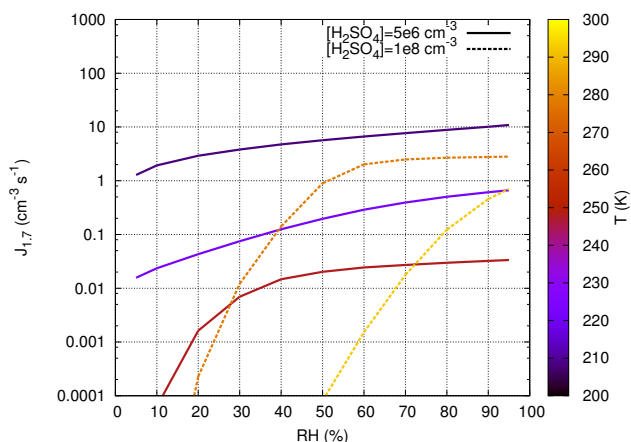


Figure S2. Modelled particle formation rate as function of relative humidity under GCR conditions in CLOUD ($4 \text{ ion pairs cm}^{-3} \text{ s}^{-1}$). The solid line gives results for temperatures below 273 K while the dashed lines give results for warmer temperatures. The temperature is also given in the colour scale. For the colder temperatures the sulfuric acid concentration was $5 \cdot 10^6 \text{ cm}^3$ and for the warmer temperatures $1 \cdot 10^8 \text{ cm}^{-3}$.

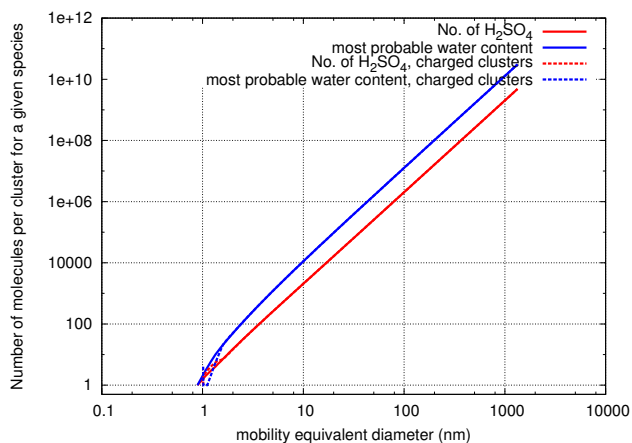


Figure S3. Number of sulphuric acid and water molecules in the particles as function of diameter for neutral particles (solid lines) and charged particles (solid lines) at 278 K and 40% relative humidity.

Table S4. Summary of SAWNUC:

Max. cluster diameter for steady state calculation	1.7 nm
Max. diameter linear size bin	3.3 nm
Geometric scaling factor	1.15
Diameter of a particle	fixed H ₂ SO ₄ and most prob- able water content
Most recent description of :	
neutral thermodynamics	<i>Kazil and Lovejoy</i> [2007]
charged thermodynamics	<i>Lovejoy et al.</i> [2004]
H ₂ SO ₄ uptake	<i>Lovejoy et al.</i> [2004]; <i>Kazil and Lovejoy</i> [2007]
recombination	<i>Brasseur and Chatel</i> [1983]; <i>Franchin et al.</i> [2015]
Ions :	
Negative	(H ₂ SO ₄) _i ·HSO ₄ ⁻
Positive	(H ₂ O) ₅ ·H ₃ O ⁺

References

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