

Amiloride hydrochloride methanol disolvate

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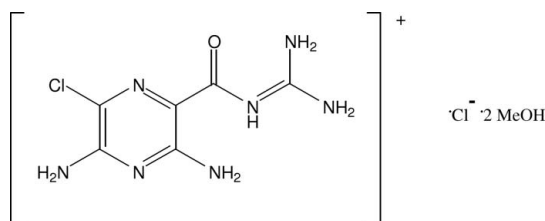
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Key indicators: single-crystal X-ray study; $T = 173$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.050; wR factor = 0.080; data-to-parameter ratio = 12.9.

In the crystal of the title compound [systematic name: 2-(3,5-diamino-6-chloropyrazin-2-ylcarbonyl)guanidinium chloride methanol disolvate], $\text{C}_6\text{H}_9\text{ClN}_7\text{O}^+\cdot\text{Cl}^-\cdot 2\text{CH}_3\text{OH}$, the components are connected by $\text{N}-\text{H}\cdots\text{N}$, $\text{N}-\text{H}\cdots\text{Cl}$, $\text{N}-\text{H}\cdots\text{O}$, $\text{O}-\text{H}\cdots\text{Cl}$ and $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds into a three-dimensional network. The dihedral angle between the aromatic ring and the guanidine residue is 6.0 (2)°.

Related literature

For other salts of amiloride, see: Pretscher *et al.* (2001); Zeslawska *et al.* (2004).



Experimental

Crystal data

$\text{C}_6\text{H}_9\text{ClN}_7\text{O}^+\cdot\text{Cl}^-\cdot 2\text{CH}_3\text{O}$
 $M_r = 330.19$
 Monoclinic, $P2_1/n$
 $a = 5.9473$ (5) Å
 $b = 16.7278$ (17) Å

$c = 14.7784$ (15) Å
 $\beta = 90.080$ (8)°
 $V = 1470.2$ (2) Å³
 $Z = 4$
 Mo $K\alpha$ radiation

$\mu = 0.46$ mm⁻¹
 $T = 173$ K

$0.30 \times 0.25 \times 0.20$ mm

Data collection

Stoe IPDS II two-circle diffractometer
 Absorption correction: multi-scan (MULABS; Spek, 2009; Blessing, 1995)
 $T_{\min} = 0.874$, $T_{\max} = 0.914$

19184 measured reflections
 2739 independent reflections
 1852 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.071$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.050$
 $wR(F^2) = 0.080$
 $S = 0.96$
 2739 reflections
 212 parameters
 9 restraints

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.22$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.25$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

| $D-\text{H}\cdots A$ | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|-------------------------------------------------------|--------------|--------------------|-------------|----------------------|
| $\text{N31}-\text{H31A}\cdots\text{N4}^{\text{i}}$ | 0.88 (1) | 2.14 (1) | 2.996 (3) | 165 (3) |
| $\text{N31}-\text{H31B}\cdots\text{Cl1}^{\text{ii}}$ | 0.87 (1) | 2.50 (2) | 3.281 (3) | 150 (3) |
| $\text{N51}-\text{H51A}\cdots\text{Cl1}^{\text{iii}}$ | 0.88 (1) | 2.54 (1) | 3.396 (3) | 165 (3) |
| $\text{N51}-\text{H51B}\cdots\text{O11}$ | 0.88 (1) | 2.11 (2) | 2.781 (3) | 133 (3) |
| $\text{N12}-\text{H12}\cdots\text{O2M}^{\text{iv}}$ | 0.87 (1) | 2.14 (2) | 2.912 (3) | 149 (3) |
| $\text{N14}-\text{H14A}\cdots\text{Cl1}^{\text{v}}$ | 0.88 (1) | 2.34 (1) | 3.188 (3) | 162 (3) |
| $\text{N14}-\text{H14B}\cdots\text{O2M}^{\text{v}}$ | 0.88 (1) | 1.93 (2) | 2.783 (3) | 162 (3) |
| $\text{N15}-\text{H15A}\cdots\text{Cl1}^{\text{v}}$ | 0.88 (1) | 2.61 (2) | 3.367 (3) | 145 (3) |
| $\text{N15}-\text{H15B}\cdots\text{O11}$ | 0.88 (1) | 2.03 (3) | 2.688 (3) | 131 (3) |
| $\text{O1M}-\text{H1M}\cdots\text{Cl1}$ | 0.84 | 2.26 | 3.091 (2) | 171 |
| $\text{O2M}-\text{H2M}\cdots\text{O1M}$ | 0.84 | 1.91 | 2.745 (3) | 170 |

Symmetry codes: (i) $-x + 2, -y, -z + 1$; (ii) $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$; (iii) $x + \frac{1}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$; (iv) $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$; (v) $x - 1, y, z$.

Data collection: *X-AREA* (Stoe & Cie, 2001); cell refinement: *X-AREA*; data reduction: *X-AREA*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *XP* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NG2772).

References

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supplementary materials

Acta Cryst. (2010). E66, o1364 [doi:10.1107/S1600536810017484]

Amiloride hydrochloride methanol disolvate

C. Q. Ton and M. Bolte

Experimental

Crystals of the title structure were obtained by recrystallization of amiloride hydrochloride (100 mg) from a methanol (3 g) solution.

Refinement

H atoms bonded to O and C were geometrically positioned and refined using a riding model with fixed individual displacement parameters [$U(\text{H}) = 1.5 U_{\text{eq}}(\text{C}, \text{O})$] using a riding model with $\text{C—H} = 0.98 \text{ \AA}$ and $\text{O—H} = 0.84 \text{ \AA}$, respectively. The methyl and hydroxyl groups were allowed to rotate but not to tip. H atoms bonded to N were refined with a distance restraint of $0.88 (1) \text{ \AA}$ and with $U(\text{H}) = 1.2 U_{\text{eq}}(\text{N})$.

Figures

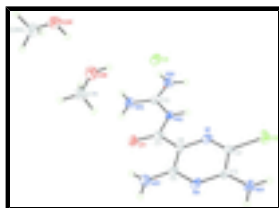


Fig. 1. Perspective view of the asymmetric unit of the title compound. Displacement ellipsoids are at the 50% probability level.

2-(3,5-diamino-6-chloropyrazin-2-ylcarbonyl)guanidinium chloride methanol disolvate

Crystal data

$\text{C}_6\text{H}_9\text{ClN}_7\text{O}^+\cdot\text{Cl}^-\cdot 2\text{CH}_4\text{O}$

$M_r = 330.19$

Monoclinic, $P2_1/n$

Hall symbol: $-P 2_1n$

$a = 5.9473 (5) \text{ \AA}$

$b = 16.7278 (17) \text{ \AA}$

$c = 14.7784 (15) \text{ \AA}$

$\beta = 90.080 (8)^\circ$

$V = 1470.2 (2) \text{ \AA}^3$

$Z = 4$

$F(000) = 688$

$D_x = 1.492 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 11166 reflections

$\theta = 3.7\text{--}25.7^\circ$

$\mu = 0.46 \text{ mm}^{-1}$

$T = 173 \text{ K}$

Block, colourless

$0.30 \times 0.25 \times 0.20 \text{ mm}$

Data collection

Stoe IPDS II two-circle

2739 independent reflections

supplementary materials

diffractometer

Radiation source: fine-focus sealed tube

graphite

ω scans

Absorption correction: multi-scan
(MULABS; Spek, 2009; Blessing, 1995)

$T_{\min} = 0.874$, $T_{\max} = 0.914$

19184 measured reflections

1852 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.071$

$\theta_{\max} = 25.6^\circ$, $\theta_{\min} = 3.6^\circ$

$h = -7 \rightarrow 7$

$k = -20 \rightarrow 20$

$l = -17 \rightarrow 17$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.050$

$wR(F^2) = 0.080$

$S = 0.96$

2739 reflections

212 parameters

9 restraints

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0253P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.22 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.25 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|--------------|---------------|--------------|----------------------------------|
| N1 | 0.4136 (4) | 0.07364 (14) | 0.32999 (16) | 0.0187 (6) |
| C2 | 0.5851 (5) | 0.02607 (17) | 0.32413 (18) | 0.0179 (6) |
| Cl2 | 0.62266 (14) | -0.02883 (5) | 0.22493 (5) | 0.0315 (2) |
| C3 | 0.7485 (5) | 0.01732 (16) | 0.39568 (19) | 0.0182 (6) |
| N31 | 0.9232 (4) | -0.03176 (16) | 0.38692 (17) | 0.0241 (6) |
| H31A | 1.006 (5) | -0.0400 (18) | 0.4354 (14) | 0.029* |
| H31B | 0.939 (5) | -0.0621 (16) | 0.3394 (14) | 0.029* |
| N4 | 0.7256 (4) | 0.06106 (13) | 0.47133 (16) | 0.0179 (5) |
| C5 | 0.5473 (5) | 0.11111 (16) | 0.47856 (19) | 0.0173 (6) |

| | | | | |
|------|--------------|--------------|--------------|--------------|
| N51 | 0.5303 (5) | 0.15414 (15) | 0.55477 (18) | 0.0256 (6) |
| H51A | 0.624 (4) | 0.1497 (18) | 0.6007 (15) | 0.031* |
| H51B | 0.410 (3) | 0.1839 (16) | 0.561 (2) | 0.031* |
| C6 | 0.3874 (5) | 0.11725 (17) | 0.40695 (19) | 0.0168 (6) |
| C11 | 0.1930 (5) | 0.17089 (18) | 0.4093 (2) | 0.0202 (7) |
| O11 | 0.1433 (4) | 0.21444 (12) | 0.47391 (14) | 0.0249 (5) |
| N12 | 0.0642 (4) | 0.16832 (14) | 0.33015 (15) | 0.0173 (5) |
| H12 | 0.110 (5) | 0.1370 (14) | 0.2874 (15) | 0.021* |
| C13 | -0.1158 (5) | 0.21632 (17) | 0.3097 (2) | 0.0196 (7) |
| N14 | -0.2054 (4) | 0.20867 (16) | 0.22820 (18) | 0.0250 (6) |
| H14A | -0.326 (3) | 0.2377 (16) | 0.218 (2) | 0.030* |
| H14B | -0.134 (5) | 0.1739 (15) | 0.1935 (18) | 0.030* |
| N15 | -0.1985 (4) | 0.26602 (16) | 0.37008 (18) | 0.0250 (6) |
| H15A | -0.310 (4) | 0.2979 (16) | 0.355 (2) | 0.030* |
| H15B | -0.141 (5) | 0.2664 (19) | 0.4246 (11) | 0.030* |
| Cl1 | 0.36907 (12) | 0.32420 (4) | 0.23835 (5) | 0.02329 (18) |
| O1M | 0.4195 (4) | 0.44145 (13) | 0.39875 (15) | 0.0335 (6) |
| H1M | 0.4001 | 0.4139 | 0.3519 | 0.050* |
| C1M | 0.2507 (6) | 0.4216 (2) | 0.4641 (2) | 0.0355 (8) |
| H1M1 | 0.2637 | 0.3650 | 0.4802 | 0.053* |
| H1M2 | 0.1015 | 0.4318 | 0.4384 | 0.053* |
| H1M3 | 0.2716 | 0.4545 | 0.5183 | 0.053* |
| O2M | 0.4108 (4) | 0.59993 (13) | 0.35076 (15) | 0.0315 (6) |
| H2M | 0.4196 | 0.5504 | 0.3592 | 0.047* |
| C2MA | 0.2302 (6) | 0.6318 (2) | 0.4037 (2) | 0.0370 (9) |
| H2M1 | 0.0883 | 0.6071 | 0.3848 | 0.056* |
| H2M2 | 0.2214 | 0.6898 | 0.3945 | 0.056* |
| H2M3 | 0.2573 | 0.6205 | 0.4678 | 0.056* |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| N1 | 0.0148 (14) | 0.0184 (13) | 0.0229 (14) | -0.0003 (11) | -0.0025 (10) | 0.0018 (11) |
| C2 | 0.0171 (15) | 0.0209 (15) | 0.0157 (14) | -0.0018 (13) | -0.0030 (12) | 0.0000 (12) |
| Cl2 | 0.0278 (5) | 0.0391 (5) | 0.0276 (4) | 0.0122 (4) | -0.0061 (3) | -0.0114 (4) |
| C3 | 0.0142 (15) | 0.0149 (15) | 0.0254 (16) | -0.0011 (13) | 0.0002 (12) | 0.0037 (13) |
| N31 | 0.0209 (14) | 0.0244 (15) | 0.0268 (15) | 0.0082 (12) | -0.0059 (12) | -0.0050 (12) |
| N4 | 0.0133 (13) | 0.0146 (12) | 0.0258 (14) | 0.0018 (10) | -0.0036 (10) | 0.0012 (10) |
| C5 | 0.0172 (16) | 0.0129 (14) | 0.0220 (16) | -0.0037 (12) | -0.0017 (12) | 0.0012 (12) |
| N51 | 0.0250 (16) | 0.0269 (15) | 0.0250 (15) | 0.0067 (12) | -0.0070 (12) | -0.0061 (12) |
| C6 | 0.0125 (15) | 0.0141 (14) | 0.0238 (16) | -0.0012 (12) | -0.0014 (12) | -0.0002 (12) |
| C11 | 0.0186 (16) | 0.0162 (15) | 0.0257 (16) | -0.0046 (13) | -0.0006 (13) | 0.0037 (14) |
| O11 | 0.0265 (12) | 0.0212 (11) | 0.0268 (12) | 0.0062 (9) | -0.0021 (9) | -0.0042 (9) |
| N12 | 0.0174 (13) | 0.0166 (13) | 0.0180 (13) | 0.0043 (11) | -0.0024 (10) | -0.0011 (10) |
| C13 | 0.0141 (15) | 0.0156 (15) | 0.0290 (17) | -0.0004 (12) | 0.0001 (13) | 0.0027 (13) |
| N14 | 0.0197 (15) | 0.0272 (15) | 0.0280 (16) | 0.0083 (11) | -0.0061 (12) | 0.0000 (12) |
| N15 | 0.0197 (15) | 0.0263 (15) | 0.0290 (15) | 0.0103 (12) | -0.0039 (12) | -0.0018 (12) |
| Cl1 | 0.0196 (4) | 0.0206 (4) | 0.0296 (4) | 0.0024 (3) | -0.0022 (3) | 0.0006 (3) |

supplementary materials

| | | | | | | |
|------|-------------|-------------|-------------|--------------|--------------|--------------|
| O1M | 0.0372 (14) | 0.0317 (14) | 0.0317 (13) | -0.0091 (11) | 0.0034 (11) | -0.0026 (10) |
| C1M | 0.040 (2) | 0.0347 (19) | 0.0318 (19) | -0.0092 (17) | -0.0015 (16) | 0.0054 (16) |
| O2M | 0.0344 (14) | 0.0271 (13) | 0.0328 (13) | -0.0055 (11) | -0.0006 (11) | 0.0064 (11) |
| C2MA | 0.039 (2) | 0.033 (2) | 0.039 (2) | -0.0027 (17) | 0.0032 (17) | 0.0001 (16) |

Geometric parameters (Å, °)

| | | | |
|---------------|------------|----------------|------------|
| N1—C2 | 1.297 (4) | N12—H12 | 0.866 (10) |
| N1—C6 | 1.360 (4) | C13—N15 | 1.316 (4) |
| C2—C3 | 1.443 (4) | C13—N14 | 1.322 (4) |
| C2—C12 | 1.745 (3) | N14—H14A | 0.880 (10) |
| C3—N31 | 1.331 (4) | N14—H14B | 0.884 (10) |
| C3—N4 | 1.343 (4) | N15—H15A | 0.880 (10) |
| N31—H31A | 0.880 (10) | N15—H15B | 0.876 (10) |
| N31—H31B | 0.872 (10) | O1M—C1M | 1.432 (4) |
| N4—C5 | 1.355 (4) | O1M—H1M | 0.8400 |
| C5—N51 | 1.341 (4) | C1M—H1M1 | 0.9800 |
| C5—C6 | 1.426 (4) | C1M—H1M2 | 0.9800 |
| N51—H51A | 0.879 (10) | C1M—H1M3 | 0.9800 |
| N51—H51B | 0.878 (10) | O2M—C2MA | 1.433 (4) |
| C6—C11 | 1.464 (4) | O2M—H2M | 0.8400 |
| C11—O11 | 1.238 (3) | C2MA—H2M1 | 0.9800 |
| C11—N12 | 1.398 (4) | C2MA—H2M2 | 0.9800 |
| N12—C13 | 1.371 (4) | C2MA—H2M3 | 0.9800 |
| C2—N1—C6 | 118.4 (2) | C11—N12—H12 | 117 (2) |
| N1—C2—C3 | 122.9 (3) | N15—C13—N14 | 121.9 (3) |
| N1—C2—C12 | 118.7 (2) | N15—C13—N12 | 120.8 (3) |
| C3—C2—C12 | 118.4 (2) | N14—C13—N12 | 117.2 (3) |
| N31—C3—N4 | 119.8 (3) | C13—N14—H14A | 116 (2) |
| N31—C3—C2 | 121.1 (3) | C13—N14—H14B | 114 (2) |
| N4—C3—C2 | 119.1 (3) | H14A—N14—H14B | 131 (3) |
| C3—N31—H31A | 117 (2) | C13—N15—H15A | 119 (2) |
| C3—N31—H31B | 122 (2) | C13—N15—H15B | 119 (2) |
| H31A—N31—H31B | 120 (3) | H15A—N15—H15B | 122 (3) |
| C3—N4—C5 | 118.8 (2) | C1M—O1M—H1M | 109.5 |
| N51—C5—N4 | 117.2 (3) | O1M—C1M—H1M1 | 109.5 |
| N51—C5—C6 | 122.3 (3) | O1M—C1M—H1M2 | 109.5 |
| N4—C5—C6 | 120.5 (3) | H1M1—C1M—H1M2 | 109.5 |
| C5—N51—H51A | 124 (2) | O1M—C1M—H1M3 | 109.5 |
| C5—N51—H51B | 117 (2) | H1M1—C1M—H1M3 | 109.5 |
| H51A—N51—H51B | 119 (3) | H1M2—C1M—H1M3 | 109.5 |
| N1—C6—C5 | 120.3 (3) | C2MA—O2M—H2M | 109.5 |
| N1—C6—C11 | 116.1 (2) | O2M—C2MA—H2M1 | 109.5 |
| C5—C6—C11 | 123.6 (3) | O2M—C2MA—H2M2 | 109.5 |
| O11—C11—N12 | 122.2 (3) | H2M1—C2MA—H2M2 | 109.5 |
| O11—C11—C6 | 124.6 (3) | O2M—C2MA—H2M3 | 109.5 |
| N12—C11—C6 | 113.2 (3) | H2M1—C2MA—H2M3 | 109.5 |
| C13—N12—C11 | 126.4 (2) | H2M2—C2MA—H2M3 | 109.5 |
| C13—N12—H12 | 116 (2) | | |

| | | | |
|---------------|------------|-----------------|------------|
| C6—N1—C2—C3 | 0.1 (4) | N51—C5—C6—N1 | 178.7 (3) |
| C6—N1—C2—Cl2 | -178.9 (2) | N4—C5—C6—N1 | -0.8 (4) |
| N1—C2—C3—N31 | -179.6 (3) | N51—C5—C6—C11 | 0.6 (4) |
| Cl2—C2—C3—N31 | -0.5 (4) | N4—C5—C6—C11 | -178.9 (3) |
| N1—C2—C3—N4 | -1.1 (4) | N1—C6—C11—O11 | 179.6 (3) |
| Cl2—C2—C3—N4 | 178.0 (2) | C5—C6—C11—O11 | -2.2 (4) |
| N31—C3—N4—C5 | 179.5 (3) | N1—C6—C11—N12 | -0.1 (4) |
| C2—C3—N4—C5 | 1.1 (4) | C5—C6—C11—N12 | 178.1 (3) |
| C3—N4—C5—N51 | -179.7 (2) | O11—C11—N12—C13 | 6.0 (4) |
| C3—N4—C5—C6 | -0.2 (4) | C6—C11—N12—C13 | -174.3 (3) |
| C2—N1—C6—C5 | 0.8 (4) | C11—N12—C13—N15 | -6.9 (4) |
| C2—N1—C6—C11 | 179.0 (3) | C11—N12—C13—N14 | 175.0 (3) |

Hydrogen-bond geometry (\AA , $^\circ$)

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--------------------------------------|----------|-------------|-------------|---------------|
| N31—H31A \cdots N4 ⁱ | 0.88 (1) | 2.14 (1) | 2.996 (3) | 165 (3) |
| N31—H31B \cdots Cl1 ⁱⁱ | 0.87 (1) | 2.50 (2) | 3.281 (3) | 150 (3) |
| N51—H51A \cdots Cl1 ⁱⁱⁱ | 0.88 (1) | 2.54 (1) | 3.396 (3) | 165 (3) |
| N51—H51B \cdots O11 | 0.88 (1) | 2.11 (2) | 2.781 (3) | 133 (3) |
| N12—H12 \cdots O2M ^{iv} | 0.87 (1) | 2.14 (2) | 2.912 (3) | 149 (3) |
| N14—H14A \cdots Cl1 ^v | 0.88 (1) | 2.34 (1) | 3.188 (3) | 162 (3) |
| N14—H14B \cdots O2M ^{iv} | 0.88 (1) | 1.93 (2) | 2.783 (3) | 162 (3) |
| N15—H15A \cdots Cl1 ^v | 0.88 (1) | 2.61 (2) | 3.367 (3) | 145 (3) |
| N15—H15B \cdots O11 | 0.88 (1) | 2.03 (3) | 2.688 (3) | 131 (3) |
| O1M—H1M \cdots Cl1 | 0.84 | 2.26 | 3.091 (2) | 171. |
| O2M—H2M \cdots O1M | 0.84 | 1.91 | 2.745 (3) | 170. |

Symmetry codes: (i) $-x+2, -y, -z+1$; (ii) $-x+3/2, y-1/2, -z+1/2$; (iii) $x+1/2, -y+1/2, z+1/2$; (iv) $-x+1/2, y-1/2, -z+1/2$; (v) $x-1, y, z$.

Fig. 1

