

Simulation of denitrification and ozone loss for the Arctic winter 2002/2003

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Abstract. We present simulations with the Chemical Lagrangian Model of the Stratosphere (CLaMS) for the Arctic winter 2002/2003. We integrated a Lagrangian denitrification scheme into the three-dimensional version of CLaMS that calculates the growth and sedimentation of nitric acid trihydrate (NAT) particles along individual particle trajectories. From those, we derive the HNO₃ downward flux resulting from different particle nucleation assumptions. The simulation results show a clear vertical redistribution of total inorganic nitrogen (NO_v), with a maximum vortex average permanent NO_v removal of over 5 ppb in late December between 500 and 550 K and a corresponding increase of NO_v of over 2 ppb below about 450 K. The simulated vertical redistribution of NO_v is compared with balloon observations by MkIV and in-situ observations from the high altitude aircraft Geophysica. Assuming a globally uniform NAT particle nucleation rate of 7.8×10^{-6} cm⁻³ h⁻¹ in the model, the observed denitrification is well reproduced.

In the investigated winter 2002/2003, the denitrification has only moderate impact (\leq 14%) on the simulated vortex average ozone loss of about 1.1 ppm near the 460 K level. At higher altitudes, above 600 K potential temperature, the simulations show significant ozone depletion through NO_xcatalytic cycles due to the unusual early exposure of vortex air to sunlight.

1 Introduction

During the Arctic winter 2002/2003, an extensive coordinated US-European measurement campaign (SOLVE II/VIN-TERSOL) was carried out to investigate a variety of open

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questions in polar stratospheric research. Within this campaign in-situ and remote measurements from balloons, high and low flying aircraft, satellite and ground stations were collected. Most measurements were performed from Kiruna, Sweden (68° N).

The winter 2002/2003 was characterized by extremely low temperatures in early winter (Naujokat and Grunow, 2003; Tilmes et al., 2003) that are necessary for polar stratospheric cloud (PSC) formation. Around 18 November, temperatures fell below the threshold temperature T_{NAT} below which NAT (Nitric Acid Trihydrate) particles can exist (Hanson and Mauersberger, 1988). In December there were recordlarge areas with temperatures below T_{NAT} . During this time PSCs were observed both from ground based LIDAR and from satellite (Spang et al., 2005). In mid-January, a major warming led to the rise of temperatures above T_{NAT} so that PSCs disappeared. In early February the vortex stabilized again and low temperatures necessary for PSC formation were present again for a few days. Finally the vortex broke down in late April.

In this study, we focus on the simulation of denitrification throughout the winter. The importance of denitrification for polar stratospheric chemistry has been known for a long time (Fahey et al., 1990; Waibel et al., 1999). However, only in the last years the processes involved in denitrification have become clearer. In-situ observations of PSCs in the Arctic winter 1999/2000 (Fahey et al., 2001) have led to the discovery of large HNO₃ containing PSC particles up to 20 μ m diameter with low number density of about 10^{-4} cm⁻³. Due to this low number density, those particles can grow to large sizes within a few days, sediment and transport significant amounts of HNO₃ downwards, consequently leading to denitrification (Carslaw et al., 2002).

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In addition to the mode with low particle number density, smaller NAT particles at higher number densities of about 10^{-3} cm⁻³ have frequently been observed in the Arctic stratosphere (Deshler et al., 2003; Larsen et al., 2004; Fahey et al., 2001). Because the settling velocity of a 3 μ m diameter NAT particle is about 90 m per day on the 500 K level, the smaller particles do not sediment significantly. Instead, the growth of other NAT particles is reduced in the presence of such higher particle number densities because those particles reduce the available HNO₃ in the gas phase due to uptake into the particles until equilibrium between particles and gas phase is reached. At particle number densities above 10^{-3} cm⁻³, the equilibrium is reached within less than one day (B. Luo, personal communication, 2004).

The vertical redistribution of HNO3 due to the sedimentation of the large NAT particles has an important impact on stratospheric chemistry. Firstly, the reactive surface and volume of the solid and liquid PSCs and indeed the threshold temperature for PSC existence depend on the available HNO₃ in the gas phase. Secondly, a loss of HNO₃ due to denitrification implies a slower increase of NO_x (=NO+NO₂) at a later time as HNO₃ is the major reservoir of NO_x. In the Arctic, the reaction of NO₂ with ClO is mainly responsible for the chlorine deactivation at the end of the PSC period. Thus, the vertical redistribution of HNO₃ influences both chlorine activation and deactivation. Model simulations that aim to reproduce ozone realistically must therefore include a realistic denitrification scheme. The nucleation process of NAT particles at low number densities has not yet been identified. Larger number densities of NAT particles could nucleate on ice in lee-wave events (Luo et al., 2003; Voigt et al., 2003; Carslaw et al., 2002). Recent observations (Voigt et al., 2005; Larsen et al., 2004) show that NAT particles also form at short timescales of about one day at temperatures above the ice frost-point. Tabazadeh et al. (2002) suggested a surface nucleation mechanism, but the nucleation rate must be more than an order of magnitude lower to explain the particle observations (Larsen et al., 2004). Studies for the winter 1999/2000 use a globally uniform, volume averaged nucleation rate independent of temperature and achieve realistic denitrification (Carslaw et al., 2002; Mann et al., 2002, 2003). The NAT nucleation rates derived from observations in the winter 2002/2003 (Voigt et al., 2005; Larsen et al., 2004) are higher compared to the uniform nucleation rate used by Carslaw et al. (2002) in their study of denitrification in the winter 1999/2000. Another approach to explain the large NAT particle mode employs a localized nucleation on ice particles at the bottom of a so-called mother cloud (Fueglistaler et al., 2002). However, since observations show that NAT particles can nucleate without the mother cloud mechanism (Voigt et al., 2005), we follow here the first assumption, i.e. a globally uniform, volume averaged nucleation rate independent of temperature.

The remaining uncertainties especially with respect to the nucleation process cannot be resolved here. The aim of this

study is to find combinations of parameters using the available knowledge of NAT nucleation that allow to simulate the observed denitrification.

2 Model description

Current chemistry transport model (CTM) simulations taking into account vertical NO_y redistribution by sedimentation of large NAT particles use a combination of a Lagrangian particle model with an Eulerian CTM (Carslaw et al., 2002; Mann et al., 2003). Here, we present simulations in which both the NAT particles and the chemistry are simulated in a Lagrangian way.

The Chemical Lagrangian Model of the Stratosphere (CLaMS) was initially developed as a two-dimensional Lagrangian chemical transport model on isentropic surfaces (McKenna et al., 2002b,a). Recently, it was further developed as a full three-dimensional model (Konopka et al., 2004). In an earlier version of the model, denitrification was parameterized in a way that only the loss of HNO₃ is described with a rate depending on the assumed size of the PSC particles (Grooß et al., 2002). Here, we use a more realistic approach that takes into account the HNO₃ flux due to sedimenting large PSC particles and also considers the HNO₃ increase due to particle evaporation.

The calculation of growth and evaporation of large NAT particles is carried out following the procedure described in detail by Carslaw et al. (2002). The growth of each particle leads to a change of its sedimentation velocity, which is computed with a time-step of 30 min. All particles (hereafter called particle boxes, to emphasize the technical character of the term and to avoid misinterpretation and confusion) are advected horizontally on isentropic levels using the background wind fields. The vertical motion is calculated using the sedimentation velocity only. The particle boxes are advected until the particles have evaporated completely. New particle boxes are nucleated every 24 h within 32 isentropic layers between 400 and 700 K north of 50° N at locations with $T < T_{NAT}$ with a mean distance of 67 km, the density of particle boxes being a factor of about 4 higher than the density of the CLaMS air parcels for which the chemistry is calculated. This was done to assure that in areas where particle boxes occur, all CLaMS air parcels have the chance to interact with at least one particle box.

Each particle box represents a much higher number of real atmospheric particles to keep the number of simulated particle boxes computationally manageable. Similar to Carslaw et al. (2002), a scaling factor in form of a number density is applied during the interaction between the particle boxes and the background chemistry. An effective nucleation rate can be calculated from the product of the number density of atmospheric particles in the newly added particle boxes and the ratio of the particle boxes and the air parcels, divided by the CLaMS time step (24 h).

The interaction between the particle boxes and the background chemistry is managed by first determining the three nearest neighboring air parcels within the layers above and below the current position of every particle box, thus creating a polyhedron containing the particle box. Nitric acid and water vapor being taken up or released during growth or evaporation, respectively, are then taken from or added to these six nearest air parcels using a distance depending weight. The impact of the interaction between particle boxes and air parcels is then determined by assigning a particle number density (the above mentioned scaling factor) to each particle box to compute the exchange of nitric acid and water vapor.

In this context it should be noted, that the density of the air parcels within each isentropic layer are not conserved during the simulation due to the properties of the mixing algorithm. This was also taken into account when calculating the mass exchange between CLaMS air parcels and particle boxes by applying an appropriate correction factor.

In sensitivity simulations, we changed the NAT nucleation rate in the model to investigate its impact on the resulting denitrification. In the standard case, we use a constant effective nucleation rate of 7.8×10^{-6} cm⁻³ h⁻¹ at all locations where T < T_{NAT}, because this results in the best comparison with the observations (see below). This is about the same as the rate reported by Voigt et al. (2005) and a factor of 3.2 lower than the rate suggested by Larsen et al. (2004). The nucleation rate is about a factor of 2.7 larger than used by Carslaw et al. (2002) for the winter 1999/2000.

The mode with high particle number density was not generated by the sedimentation scheme. To achieve this mode we used the microphysics module already implemented in CLaMS (Carslaw et al., 1995; McKenna et al., 2002a), in which a particle mode with given number density (here: $0.003 \,\mathrm{cm}^{-3}$) can nucleate upon cooling from liquid aerosol at a certain supersaturation with respect to NAT or from sulfuric acid tetrahydrate (SAT). This critical supersaturation of HNO₃ over NAT was varied and was set to 30 in the reference simulation corresponding to a temperature about 5 K below T_{NAT}. NAT formation from SAT is possible in this scheme but is not allowed in the reference simulation. Sedimentation of this mode is not considered. Therefore there are three possibilities to change the denitrification parameterization: A change of the rate of nucleation for the mode with low particle density, a change of the critical supersaturation to form the mode with high particle density, or a change whether or not NAT particles can form on SAT upon cooling. The SAT particles in the last case originate from previously evaporated NAT particles.

In the current simulation we use a horizontal resolution of 100 km in the area north of 40° N and 300 km between the equator and 40° N for the air parcels. The vertical model domain is between 350 and 900 K potential temperature divided into 30 levels.

3 Initialization and boundary conditions

The presented CLaMS simulation was started for 17 November 2002 to ensure that it starts before the onset of PSC formation that is also the begin of denitrification. The chemical composition of the air parcels in the model domain was derived differently for the individual chemical species depending on the available data.

The initialization of O3 was based on the MIPAS-ENVISAT data from 16 and 17 November 2002. Here, we used the near-real-time data version of the MIPAS data provided by the European Space Agency (ESA). Data profiles for which the retrieval is potentially disturbed due to PSCs are excluded from the data set using the cloud index defined by Spang et al. (2005). From the observation locations and times trajectories on 19 isentropic levels between 350 and 2000 K were started to determine the locations of the observed air masses at the synoptic time 17 November 12:00 UT. These data then were gridded onto a regular $2^{\circ} \times 6^{\circ}$ latitude-longitude grid on each level. Points within 440 km distance of a grid point contributed to it with different weights taking into account its distance and its given relative accuracy. CH₄ was initialized using HALOE satellite and MkIV balloon data. Since HALOE observations (Russell et al., 1993) are available much less frequently than MIPAS data, a different method and a longer time interval was used. The HALOE CH₄ observations of November and December 2002 were averaged into equivalent latitude bins on the different isentropic levels. Since HALOE barely sampled the vortex during this time the data were combined with vortex CH₄ observations on 16 December 2002 obtained by MkIV (Toon et al., 1999). Correlations of CH₄ with other compounds (N₂O, Cl_v , Br_v) were used as in Grooß et al. (2002), with the exception that Bry was increased by 10% to maximum mixing ratios of about 20 ppt. Total reactive nitrogen (NO_v) was initialized using the correlation with N₂O given by Popp et al. (2001). For air parcels with N_2O mixing ratios between 1 and 33 ppb the correlation was extended here on the basis of the same data (MkIV balloon profile in December 1999) using the following relation:

$$[NO_y] = 11.7 + 4.49 \times \ln [N_2O] - 0.948 \times (\ln [N_2O])^2, (1)$$

where NO_v and N_2O mixing ratios are given in ppb.

The sulfate background aerosol in the model is given as H_2SO_4 equivalent gas phase mixing ratio. Its altitude dependence (given in ppb) was derived from a detailed study (Bausch et al., manuscript in preparation) to be

$$[H_2SO_4] = 1.21 \times 10^{-8} \theta^3 - 3.67 \times 10^{-5} \theta^2 +2.96 \times 10^{-2} \theta - 7.97,$$
(2)

for the potential temperature range $450 \text{ K} \le \theta \le 900 \text{ K}$. At the upper boundary of the model domain (900 K), also the chemical composition of the air parcels was prescribed derived in a similar way from the satellite data. At the lower boundary (350 K), the chemical composition of the air parcels was



Fig. 1. Simulated average size distribution of low particle number density mode on 24 December 2002. Shown is the distribution of particle radius R for the area of NAT particle existence converted to dN/d log R for different potential temperature ranges as indicated in the legend.

not changed during the run. This was done because there is general downward motion in the vortex over the winter and with this method the airmasses near the lower boundary stay self-consistent.

Similar as the initialization, the boundary conditions at the upper boundary of the model domain were determined. At the upper boundary (900 K), ENVISAT-MIPAS data for O_3 and CH₄ were averaged over equivalent latitude bins for the simulation time every half month within 5 days. From that data, the same correlations as above were used to determine N₂O, Cl_y, Br_y, and NO_y. Species not mentioned and the partitioning within the families were taken from the Mainz 2-D photochemical model (Grooß, 1996).

The ozone mixing ratio needed for the radiative transfer code (Becker et al., 2000) that determines the photolysis rates has been taken from a climatology based on HALOE measurements as function of latitude, pressure and month (Grooß and Russell, 2005).

4 Denitrification

For the current simulation, the downward NO_y flux was very variable depending on temperature. The maximum number of about 145 000 particle boxes was simulated on 24 December. To derive a size distribution of NAT particles from the model results one has to average over a large volume and has to weight it by the ratio of present particle boxes to air parcels. The simulated particle size distribution for 24 December is shown in Fig. 1 for the whole area where NAT particles exist divided into different potential temperature regions. It is evident that at the lower levels larger particles were simulated since they did have more time to grow. It is



Fig. 2. Redistribution of NO_y as function of time and potential temperature from the reference simulation $(J=7.8\times10^{-6} \text{ cm}^{-3} \text{ h}^{-1})$. Shown is the average over equivalent latitude range 65–90° N for the period 17 November 2002 to 10 February 2003. The zero contour is displayed as a black line.

also evident that with this parameterization, no high particle number density mode at low sizes is present. This is the reason why the high particle number density mode was added separately to the code.

Figure 2 shows the simulated NO_v change due to the sedimentation of the large NAT particles averaged over equivalent latitude range 65-90° N for the reference simulation. The unperturbed NO_v^* was simulated by a passive tracer that was initialized identically as NOv and that was then advected and mixed without chemistry and denitrification. The maximum denitrification of over 5 ppb is reached in early January. After that time, no significant additional denitrification was simulated and the denitrified air masses were only advected further. Also clearly visible is the re-nitrification at lower levels caused by the evaporation of the NAT particles. The diabatic descent within the polar vortex causes the downward motion of the denitrification peak. The mixing ratio increase at the re-nitrification altitudes is lower than the mixing ratio decrease at denitrification altitudes, because the total mass of sedimented HNO₃ should be conserved and the pressure increases with decreasing altitude. Visible is also the very small chemical production of NO_v through the reaction of N_2O with $O(^1D)$ in early February above 600 K.

Figure 3 shows examples for simulated NO_y change on 19 January for the 425 K and 450 K levels. Note that the simulated NO_y redistribution is very inhomogeneous in space at that time. This is due to the temperature threshold of the process. At locations with temperatures just above T_{NAT} , all NAT particles evaporate relatively fast, which leads to strong re-nitrification, while possibly particles at close by locations with slightly lower temperatures can sediment further down.

In the following we show comparisons of the simulated denitrification with observations from MkIV and the Geophysica.



Fig. 3. Simulated redistribution of NO_y on 19 January 2003 for the potential temperature levels 425 K (left) and 450 K (right). The pink line indicates the Geophysica flight path (transformed to synoptic 12:00 UT locations). The black line marks the vortex edge.

4.1 Early denitrification observed by MkIV

On 16 December 2002, the MkIV experiment on board the OMS remote balloon gondola measured N2O and all major NO_v compounds among many other species (Toon et al., 1999). At that time, already an onset of denitrification was both observed and simulated. These data are an important constraint for the denitrification parameterization. Figures 4 and 5 show the observations of NO_v (green line) and NO_v^{*} derived from N₂O (green dotted line) using the correlation mentioned above. The deviation between NO_v and NO_v^* from MkIV between about 500 and 650 K is very likely caused by denitrification. The colored lines show the corresponding CLaMS NOv mixing ratios for different parameterizations of denitrification. In Fig. 4 the nucleation rate of sedimenting the NAT particles was varied from 3.3 to $16 \times 10^{-6} \text{ cm}^{-3} \text{ h}^{-1}$. These results show clearly that an increase of the nucleation rate yields larger denitrification. For the reference simulation (marked blue in all figures unless otherwise stated) the nucleation rate of 7.8×10^{-6} cm⁻³ h⁻¹ yields the best agreement with observed NO_v above about 500 K. At 500 K and below, both larger denitrification and less re-nitrification than observed was simulated. A nucleation rate of 3.3×10^{-6} cm⁻³ h⁻¹ as was used by Carslaw et al. (2002) yields too low denitrification in our simulation (Fig. 4, light blue line).

Additionally, the dependence of denitrification on the NAT nucleation rate was investigated for the whole vortex area over the winter 2002/2003 (not shown here). The maximum difference of simulated NO_y mixing ratio between simulations with the nucleation rates 7.8×10^{-6} and 3.3×10^{-6} cm⁻³ h⁻¹ averaged over equivalent latitudes

above 65° N was found on 16 December (1.4 ppb at about 550 K). This indicates that the sensitivity with respect to the nucleation rate is not special for the observation location but typical for the whole vortex area. Later in the winter, the difference between the two simulations decreases with time, e.g. the vortex average NO_v difference on 19 January peaks at 540 K being only 0.9 ppb. In Fig. 5, the conditions of formation of the mode with high NAT density are varied for a nucleation rate of 7.8×10^{-6} cm⁻³ h⁻¹. The red line shows a simulation in which NAT formation from liquid aerosol is completely suppressed, the blue dotted and the orange line show simulations in which the mode with high NAT density is formed at supersaturation 30 and 10, respectively. These three simulations show almost identical denitrification at the location of the MkIV observation. This is most likely the case because supersaturations of 10 and higher are probably not reached in the history of the observed air masses.

Alternatively, the observed denitrification could be explained by using a higher nucleation rates in combination with hindering the growth of the NAT particles by the presence of a NAT mode with high particle density. This mode has been observed in December 2002 by balloon and satellite measurements (Larsen et al., 2004; Spang et al., 2005). As shown in Fig. 5, this mode cannot be obtained in this case by varying the NAT supersaturation necessary to form NAT, but it can be reached by allowing NAT formation on SAT particles.

The light blue line Fig. 5 shows a simulation with the same nucleation rate $(7.8 \times 10^{-6} \text{ cm}^{-3} \text{ h}^{-1})$ in which the formation of NAT from SAT is allowed. The SAT particles in the simulation originate from previously evaporated NAT particles that remained below the melting temperature of about 215 K.



Fig. 4. NO_y observation and simulations. Shown are MkIV observation of NO_y (green line) and NO_y^{*} derived from N₂O (dotted green line) as well as CLaMS simulations for NO_y^{*} (black dotted line) and NO_y (colored lines) with different nucleation rates *J* for the mode with low particle density as indicated in the legend. The CLaMS simulations are vertically averaged over the ± 1 km vertically sampled volume by MkIV.



Fig. 5. As Fig. 4, but for different denitrification parameterizations. The colored lines show results of CLaMS simulations for nucleation rate of $J=7.8 \times 10^{-6}$ cm⁻³ h⁻¹ and different formation assumptions for the NAT mode with high particle density. No NAT formation from liquid aerosol (red), Formation at HNO₃ supersaturation over NAT of 30 (blue dotted, reference simulation) and 10 (orange), and with NAT formation from SAT in combination with a doubled nucleation rate (light blue). The black line shows a simulations as the reference simulation but with temperatures artificially increased by 1 K.

This case yielded less denitrification, comparable to a simulation with about half the nucleation rate (compare Fig. 4). However, laboratory studies show that NAT nucleation on SAT is unlikely (Iraci et al., 1995; Koop et al., 1997). But similar results should be expected if instead of SAT other yet

unknown condensation nuclei would form NAT. Therefore, the assumption of NAT nucleation on SAT is not considered further. The sensitivity with respect to temperature was also investigated. The black line in Fig. 5 shows the denitrification simulated with an artificial 1 K increase of the temperature in the ECMWF analysis. This of course yields less denitrification at the location of the MkIV observation. The largest impact, about 2 ppb less denitrification is seen around 500 K. This denitrification change is comparable to a nucleation rate decrease by a factor of two.

At about 480 K potential temperature, the MkIV observations suggest re-nitrification that seems not to be represented in the simulation. Figure 6a shows the simulated NO_y change on the 480 K level. Here, the air in the vortex core is denitrified. Around this denitrified area, a filament of re-nitrified air is visible very close to the location of the MkIV observation (pink circle). Figure 6b shows an enlargement of the area around the observed airmass on this level reduced to the vertical resolution of MkIV and the sampled MkIV volume (1 km radius, 300 km length, transformed to its 12:00 UT location) indicated by the pink line. The reason for the discrepancy could be either due to model resolution, uncertainties of wind and temperature analyses or variability within the volume sampled by MkIV. This cannot be examined in detail here.

4.2 Comparison with Geophysica observations

Between 15 January and 9 February, stratospheric in-situ measurements were obtained on 9 flights of the Geophysica aircraft within the scope of the VINTERSOL-EUPLEX campaign. Observations of chemically long-lived tracers can be used to validate the transport of the simulation. Figure 7 shows the comparison of HAGAR observations (Volk et al., in preparation, 2005¹) and CLaMS simulation of the tracer N₂O that can be considered as chemically inert over the winter. The left panel shows the comparison along the flight of 19 January 2003. The right panel shows a scatter plot for all N2O observations obtained between 19 January and 9 February 2003 and the corresponding CLaMS observations. There is some indication that the model over-estimates N₂O for mixing ratios below about 100 ppb, but in general this comparison shows that the simulation is able to reproduce well the advection and mixing of the air masses. The reason for the slight discrepancy below 100 ppb N₂O may be either due to uncertainties in the transport (resolution, vertical descent) or initialization.

Between 15 January and 4 February, no PSC particles are simulated by the sedimentation scheme which is expected because of the major warming in January. Therefore,

¹Volk, C. M., Werner, A., Wetter, T., Ivanova, E., Wollny, A., Ulanovsky, A., Ravegnani, F., Schlager, H., Konopka, P., and Toon, G.: Ozone loss within the 2003 Arctic vortex derived from in situ observations with the Geophysica aircraft, Atmos. Chem. Phys. Discuss., in preparation, 2005.



Fig. 6. Simulated redistribution of NO_y on 16 December 2002 for the potential temperature level 480 K for CLaMS simulation with 100 km horizontal resolution. The left panel shows a polar stereographic projection. White areas correspond to values less than -10 ppb. The pink circle indicates the MkIV observation point (transformed to synoptic 12:00 UT location). The black line marks the vortex edge using the vortex definition of Nash et al. (1996). The right panel shows an enlargement around the observed airmass reduced to the vertical resolution of MkIV. The volume sampled by MkIV (1 km radius and 300 km length) is indicated by the pink line.



Fig. 7. Comparison of CLaMS simulations with HAGAR N_2O observations. Time series for the Geophysica flight on 19 January 2003 (left). The right panel shows a one to one comparison for all flights between 19 January and 9 February 2003. The potential temperature of the observation is marked as indicated by the color bar.

no additional denitrification was simulated during this time period, but the denitrified air masses were advected. We compare the simulated denitrification with observations of denitrification determined from the difference of NO_y measured by the SIOUX instrument (Schlager et al., in preparation, 2005^2 ; Voigt et al., 2005) and NO_y^{*} derived from the

HAGAR N₂O measurements shown above without having to consider HNO₃ in the particle phase at these temperatures. Figure 8 shows in the top panel the altitude dependence of vertical NO_y redistribution for the parts of the flight of 19 January within the polar vortex (equivalent latitude >70°). The shape of the denitrification and re-nitrification profile is reproduced well. The absolute values are underestimated for both nucleation rates $J=3.3\times10^{-6}$ cm⁻³ h⁻¹ and for $J=7.8\times10^{-6}$ cm⁻³ h⁻¹. The larger nucleation rate results in stronger denitrification and renitrification and the

²Schlager, H., Voigt, C., Volk, M., Davies, S., Carslaw, K., Konopka, P., Roiger, A., and Stock, P.: Observations of denitrification and renitrification in the 2002–2003 Arctic winter stratosphere, Atmos. Chem. Phys. Discuss., in preparation, 2005.



Fig. 8. Comparison of denitrification from CLaMS simulation with that derived from SIOUX and HAGAR observations. The lower panels show the altitude profile within the polar vortex core (equivalent latitude >70°). Only the CLaMS data directly corresponding to the observations are shown. Blue circles correspond to the reference simulation with the nucleation rate $J=7.8 \times 10^{-6}$ cm⁻³ h⁻¹ and the pink open circles to the simultion with the nucleation rate $J=3.3 \times 10^{-6}$ cm⁻³ h⁻¹. The middle and bottom panels show the one to one comparison for all flights between 19 January and 9 February for these two nucleation rates. The potential temperature of the observation is marked as indicated by the color bar.

results agrees better with the observations. Both, observations and the simulation do show the turnover level from denitrification to re-nitrification at about 440 K at that time. The northward part of the flight on 19 January was flown



Fig. 9. Comparison of NO_y-NO_y^{*} simulated by CLaMS using a nucleation rate of $J=7.8\times10^{-6}$ cm⁻³ h⁻¹ and derived from SIOUX and HAGAR observations on Geophysica on 8 and 12 March 2003 for equivalent latitudes above 60° N. The color of the symbols denotes the equivalent latitude of the observation location.

around that 440 K level. At this level CLaMS underestimates the re-nitrification.

The bottom two panels of Fig. 8 show the comparison for all observations between 19 January and 9 February 2003 for both choices of the nucleation rate. The results are similar for both nucleation rates. While for both cases the renitrification is under-estimated by the simulation, the reference simulation with $J=7.8\times10^{-6}$ cm⁻³ h⁻¹ does agree better with the observations of denitrified air masses above about 440 K. The reason for the similar results using the different nucleation rates is a saturation effect. Since less HNO₃ is available in denitrified air, the threshold temperature for NAT formation is lowered and therefore additional denitrification is hindered. Therefore, the simulated vertical NO_y redistribution depends significantly on the assumed particle nucleation rate most notably in early winter shortly after the onset of PSCs.

Also, the reference simulation was continued until mid-March. In March, the denitrification is still well simulated. Figure 9 shows the comparison of simulated NO_y-NO_y^{*} with denitrification derived from SIOUX and HA-GAR for the flights on 8 March and 12 March for equivalent latitudes poleward 60° N during the ENVISAT validation campaign. Here there is still reasonable agreement with the hint of a larger denitrification and lower re-nitrification in the simulation. The average difference of NO_y-NO_y^{*} ($\pm 1\sigma$) between simulation and observations in March is -0.6 ± 0.6 ppb. Unlike the comparison in January and early February, the simulation with the lower nucleation rate $J=3.3\times10^{-6}$ cm⁻³ h⁻¹ does agree better with the observations in March (-0.2 ± 0.7 ppb, not shown).



Fig. 10. Simulated ozone loss as function of time and potential temperature. Shown is the vortex average using the vortex definition of Nash et al. (1996) for the period 17 November 2002 to 16 March 2003.



Fig. 11. Simulated vortex average column ozone loss between 380 and 550 K potential temperature. The blue line corresponds to the reference simulation. The green dotted line corresponds to a simulations without sedimentation. The red dashed line corresponds to a simulation in which the absorption cross sections of Cl_2O_2 are taken from Burkholder et al. (1990).

5 Simulated ozone depletion

To investigate the ozone depletion in the winter 2002/2003, the simulation was continued until mid-March in order to include the second period of low stratospheric temperatures in mid-February. The chemical ozone depletion was determined by the difference between model ozone and an artificial passive ozone tracer. This tracer was initialized on 17 November identically as ozone, was advected and mixed as all other chemical species but did not undergo any chemical change. The simulated ozone depletion averaged over the vortex as defined by Nash et al. (1996) as function of time and potential temperature is shown in Fig. 10. The simulations indicate that vortex average ozone depletion reaches about 1.1 ppm in mid-March around the 460 K level.



Fig. 12. Comparison of ozone mixing ratios simulated by CLaMS and observed by MIPAS for potential temperatures between 350 K and 500 K on 14 March 2003. The color of the symbols denotes the equivalent latitude of the observation location.

In Fig. 11 the simulated ozone column loss between 380 K and 550 K averaged over the vortex after Nash et al. (1996) is shown for the reference simulation (blue line) and different sensitivity simulations (see below). The simulated vortex average ozone column loss on 13 March between 380 K and 550 K was 46 Dobson units (21% of the partial ozone column), and between 350 K and 900 K it was 68 Dobson units (18%).

To verify that the ozone depletion was simulated correctly over the winter, the simulated ozone mixing ratios were compared with observations of ozone on 14 March 2003 made by MIPAS-ENVISAT. On that day, MIPAS data did have a good coverage of the polar vortex. Figure 12 shows the one to one comparison for MIPAS observations between 350 and 500 K potential temperature and for equivalent latitudes poleward of 60° N. The differences between simulation and observations can be explained by small-scale structures below the resolution of CLaMS and the accuracy of the MIPAS O₃ observations of about 10% (Blumenstock et al., 2003). The average difference $(\pm 1\sigma)$ between simulated and observed ozone mixing ratios (CLaMS-MIPAS) is -0.06 ± 0.24 ppm. The largest differences in the comparison occur at the edge and outside the vortex, for equivalent latitudes equatorward of 70° N. Figure 13 shows ozone comparisons with the FOX instrument (Schlager et al., in preparation, 2005^2) on board the Geophysica aircraft made during the Envisat validation campaign between 28 February and 16 March. Here the average difference (CLaMS-FOX) is also low -0.08 ± 0.20 ppm. This indicates that all significant ozone loss processes in the vortex have been covered by the simulation.

The magnitude of the ozone depletion is comparable with estimates derived from HALOE data using the tracer correlation method (Tilmes et al., 2003), but the altitude of the maximum ozone depletion is about 25 K higher in the simulation.



Fig. 13. Comparison of ozone mixing ratios simulated by CLaMS and observed by FOX on Geophysica between 28 February and 16 March 2003, The color of the symbols denotes the equivalent latitude of the observation location.

Goutail et al. (2005) report a slightly higher amount of ozone depletion, 23% of the total column derived from observations SAOZ UV-visible spectrometers and CTM passive ozone simulations. However, due to the different altitude range the results are not completely comparable.

To compare with the results of the SLIMCAT model (Feng et al., 2005), one has to take into account that they use absorption cross section of the photolysis of Cl_2O_2 by Burkholder et al. (1990) extrapolated to the wavelength of 450 nm instead of those recommended by Sander et al. (2003) that were used here. Therefore sensitivity simulations were carried out with the Burkholder et al. (1990) absorption cross section which leads to more ozone depletion.

Compared with the reference simulation (blue line in Fig. 11), the ozone column loss increased due to the photolysis of Cl₂O₂ by Burkholder et al. (1990) by about 2 Dobson units by mid January and did not increase significantly further to values between 2.5 and 3 Dobson units in late February and March (red dashed line in Fig. 11 compared with blue line). This corresponds to an increase of the column ozone depletion between 380 K and 550 K relative to the reference simulation of at most 11.2% on 14 January. At 15 March this corresponds to 6.1% more ozone depletion compared to the reference simulation. Feng et al. (2005) show the partial ozone column loss between 345 K and 670 K averaged for the area poleward of 65° N equivalent latitude. The time development of the simulated column ozone loss in our simulation (350 K-670 K) is very similar. Due to different dynamics and resolution, there is different variability from day to day. The average difference between the simulated ozone column losses in February (1-16 March) is 1.5±4.0 Dobson units $(3.5\pm4.1 \text{ Dobson units})$, where SLIMCAT simulates slightly lower column ozone loss. These small deviations are due to different resolution, tranport schemes and a simplified denitrification scheme in SLIMCAT (Davies et al., 2003) that was tested and optimized only for a cold winter.

Further, the impact of denitrification on the simulated ozone depletion was investigated through an additional simulation in which the sedimentation scheme was deactivated (green dotted line in Fig. 11). In general, the simulated ozone depletion is very similar to the reference simulation. The largest effect of the denitrification was found towards the end of the simulation when the deactivation of active chlorine species is hindered by less available NO_x due to denitrification. The maximum of the ozone depletion due to denitrification averaged over the vortex as defined by Nash et al. (1996) was determined for 16 March to be 149 ppb at 455 K potential temperature level corresponding to 14% of the ozone depletion of the reference simulation. The column ozone loss in the 380–550 K range attributed to denitrification is 3.1 DU (7% of the partial column).

Another striking feature is the simulated ozone depletion above the 600 K level. Unlike the levels below about 500 K where ozone loss is mainly caused by catalytic cycles involving chlorine and bromine compounds, no PSCs and no chlorine activation are simulated at these altitudes. The reason for the ozone depletion above 600 K are catalytic cycles involving NO_x. Especially in the winter 2002/2003, that was dynamically active, the time when vortex air was exposed to sunlight began earlier and was longer than in other winters. Therefore, also the typical summer ozone depletion started earlier. Such a behavior was also observed in the dynamically active 2002 Antarctic winter (Grooß et al., 2005).

The chemical ozone depletion inside the vortex at altitudes above 600 K was also found in the comparison of a high resolution CLaMS tracer simulation and MIPAS observations (Konopka et al., in preparation, 2005^3). A similar comparison as above with MIPAS data on 14 March 2003 between 500 K and 800 K yields a mean difference (CLaMS–MIPAS) of $+0.27\pm0.44$ ppm. That means the simulation slightly under-estimates the ozone depletion in that altitude range.

6 Conclusions

A Lagrangian module for single particle growth and advection was developed and integrated in the Chemical Lagrangian Model of the Stratosphere (CLaMS). With this model, the vertical redistribution of NO_y for the winter 2002/2003 was successfully simulated. The presented simulations demonstrate that choosing a nucleation rate of $J=7.8 \times 10^{-6}$ cm⁻³ h⁻¹ results in the best reproduction of the observed NO_y in mid-December 2002. For that time period shortly after the onset of PSCs, the simulated denitrification is most sensitive to the nucleation rate in the absence of the mode with high particle density. It has been also shown that

³Konopka, P., Grooß, J.-U., Müller, R., et al.: Halogen- versus NOx-induced ozone loss in the Arctic middle stratosphere during the 2002/03 winter and spring, Atmos. Chem. Phys. Discuss., in preparation, 2005.

the presence of the mode with high particle density (Larsen et al., 2004; Spang et al., 2005) decelerates the growth of the large NAT particles. Thus, in the presence of higher NAT particle number densities, the nucleation rate has to be increased to achieve similar results. Later in the winter in mid-January to early February, the simulated denitrification does not significantly depend on the nucleation rate.

Further, the simulated ozone loss over the simulation period was determined. By mid-March the vortex averaged ozone loss around 460 K was about 1.1 ppm comparable to other ozone loss estimates for this winter. The corresponding column ozone loss between 380 and 550 K was determined to be 46 Dobson units (21% of the partial ozone column). A sensitivity simulation showed that 14% of the vortex ozone loss on the 455 K level (corresponding to 7% ozone column depletion between 380 and 550 K) can be attributed to denitrification. At higher altitudes above 600 K also significant ozone loss was determined. This loss was caused by NO_x-catalyzed ozone loss cycles in vortex air masses with large exposure to sunlight.

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