

# ChemSusChem

Supporting Information

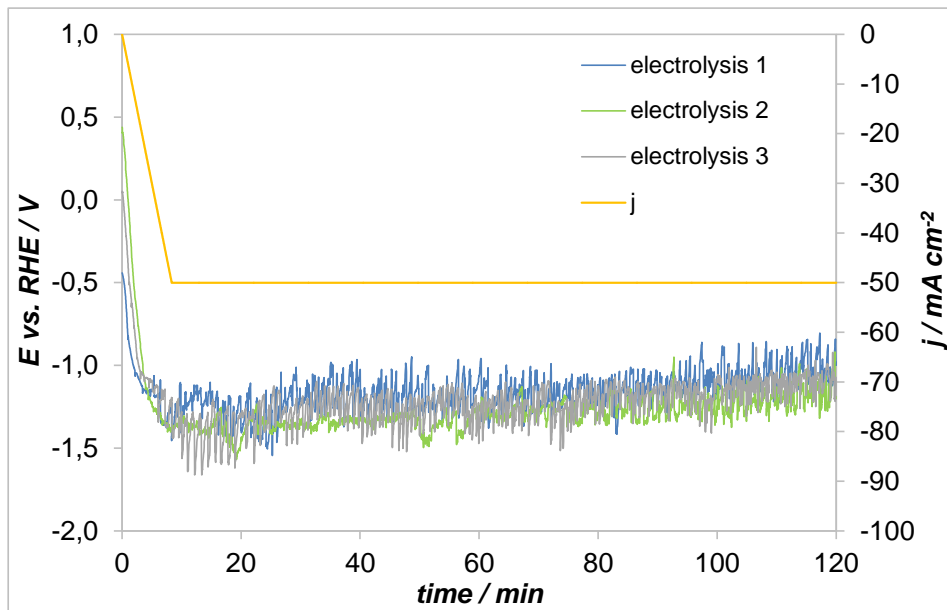
## **From CO<sub>2</sub> to Bioplastic – Coupling the Electrochemical CO<sub>2</sub> Reduction with a Microbial Product Generation by Drop-in Electrolysis**

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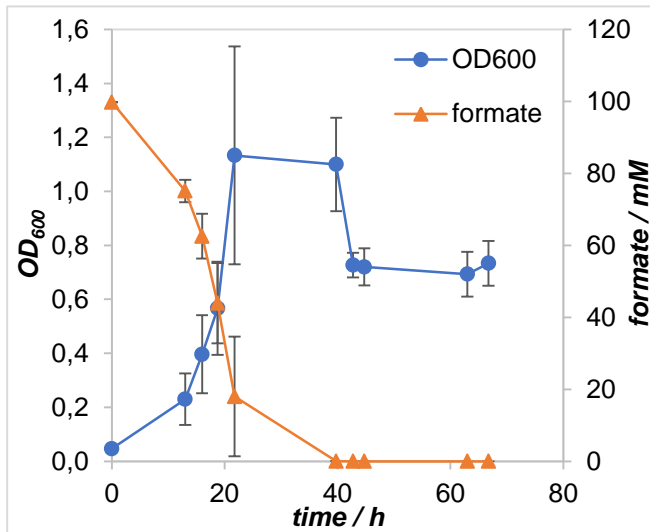
## Results and discussion

### CO<sub>2</sub> electrolysis to formate for bioconversion



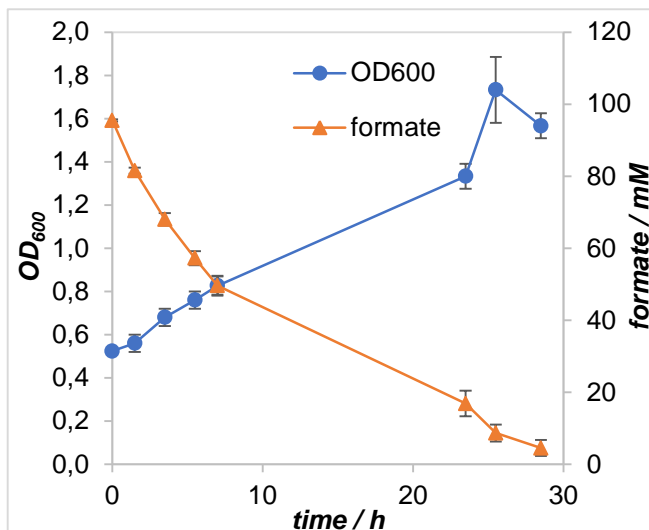
SI figure 1: Time course of electrode potential ( $E$  vs. RHE, no  $iR$ -drop correction) and current density ( $j$ ) during CO<sub>2</sub> electrolysis at Sn-based GDE in a custom designed electrolysis reactor. Blue line: electrolysis 1, green line: electrolysis 2, grey line: electrolysis 3, yellow line:  $j$ .

## Cultivation of *C. necator* on 100 mM formate medium



SI figure 2: Time course of optical density and formate concentration observed for the incubation of *C. necator* resting cells on 100 mM formate medium with manual pH control. Blue circles: OD<sub>600</sub>, orange triangles: formate concentration. Mean  $\pm$  SD, n = 3.

## PHB production with *C. necator* from formate

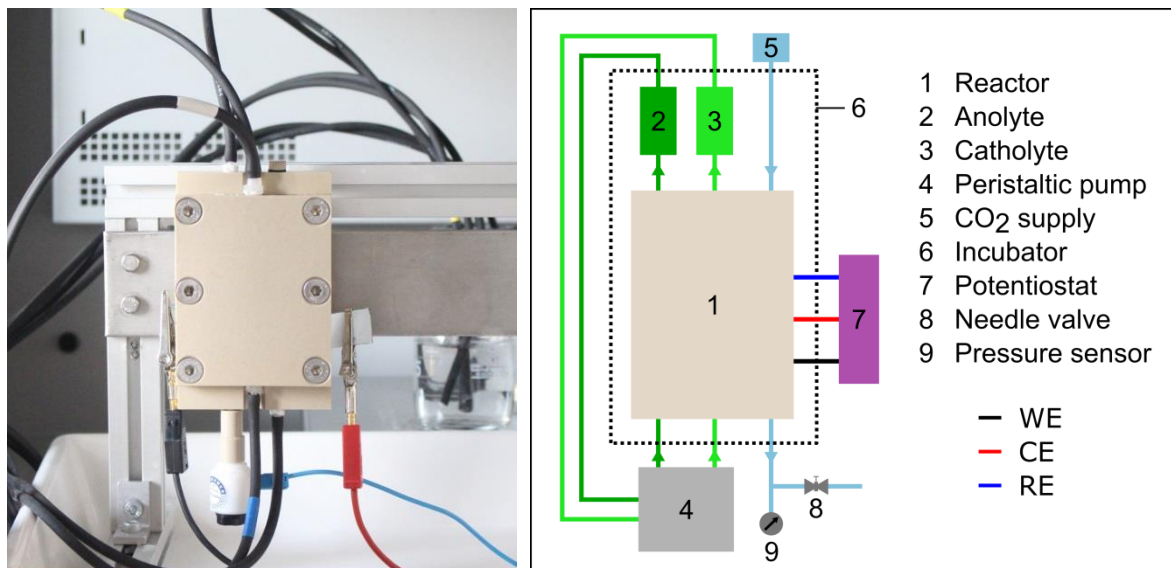


SI figure 3: Time course of optical density (OD<sub>600</sub>) and formate concentration observed for the incubation of *C. necator* resting cells on 100 mM formate medium lacking (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> with manual pH control. Blue circles: OD<sub>600</sub>, orange triangles: formate concentration. Mean  $\pm$  SD, n = 3.

## Experimental Section

### Electrochemical formate synthesis from CO<sub>2</sub>

#### Reactor set-up



SI Figure 4: Photograph (right) and schematic illustration (left) of the reactor-set-up for CO<sub>2</sub> electrolysis.